Module Handbook

Master of Science Biomedical Chemistry

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Colloid Chemistry and Medical Polymers	Complex (Supra)Molecular Systems and Biopolymers Modern Methods of Physical Chemistry Practical Course Modern Methods of Spectroscopy and Microscopy Introduction in Nuclear Chemistry Lab Course Nuclear Chemistry 1 Principles of Quantum Chemistry Programming in Quantum Chemistry Programming in Quantum Chemistry Practical Computational Chemistry Contemporary Topics of Theoretical Chemistry Artificial intelligence in drug discovery and development Remarks Abbreviations	
Modern Methods of Physical Chemistry	Modern Methods of Physical Chemistry Practical Course Modern Methods of Spectroscopy and Microscopy Introduction in Nuclear Chemistry Lab Course Nuclear Chemistry 1 Principles of Quantum Chemistry Programming in Quantum Chemistry Practical Computational Chemistry Contemporary Topics of Theoretical Chemistry Artificial intelligence in drug discovery and development Remarks Abbreviations	
Practical Course Modern Methods of Spectroscopy and Microscopy Introduction in Nuclear Chemistry Lab Course Nuclear Chemistry 1 Principles of Quantum Chemistry	Practical Course Modern Methods of Spectroscopy and Microscopy Introduction in Nuclear Chemistry Lab Course Nuclear Chemistry 1 Principles of Quantum Chemistry Programming in Quantum Chemistry Practical Computational Chemistry Contemporary Topics of Theoretical Chemistry Artificial intelligence in drug discovery and development Remarks	
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available only in German langua	bbreviations	
available only in German langua	×	
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Master of Science Biomedical Chemistry	4.0	03.04.2025	1910

Module descriptions

Mandatory Modules

Module BCF	Biocher	Biochemistry [Modul-Kennnummer]				
Mandatory or elective Module	M (2nd C	ourse as elective N	lodule)		<u>L</u>	
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration (according to course plan)	1 Semest	ter				
Courses/ Learning formats	Туре	Regel term When starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoint
a) Lecture "Molecular and Cellular Biochemistry"	L	2 (1)	Elective	4	138 h	6
or						
b) Lecture "Methods of Biochemistry"	L	1 (2)	Elective	2	69 h	3
c) Supporting seminar to b)	S	1 (2)	Elective	2 0	69 h	3
In order to complete the module, ye	ou have to	fulfil the following	requirements:			
Compulsory Attendance			(2		
Active participation	According	g to § 5 para. 3	X	V		
Coursework						
Module examination	Usually w a) or of b	ritten exam (120 m) and c)	in), alternativel	y oral exam (3)	0 min) about th	e contents of
Qualification Goals, learning outcor	ne, compe	tences				
Students are able to, a) - reproduce essential contents of cel - explain and evaluate principles of g - Evaluate the opportunities and risk their own work - Assign and explain the principles of - to understand and reproduce the b - to use relevant technical terms of c - to critically evaluate the factual kno primary literature published in inter b) and c) - to assign appropriate methods to p - to evaluate the results of bioanalyt - to understand the limitations of the - to assess the applicability of the me - to critically assess the significance of - independently acquire in-depth kno - to analyse and evaluate scientific li	gene regula s of genetic f signal trar biochemical cellular bioc owledge co national sp problems in f these me cical experir e respective ethods to n of the respective owledge of terature fro	tion and genetic en c engineering, devel asduction and cell biological chemistry correctly vered in biochemica ecialist journals the fields of protein thods ments. e methods based or ew questions. ective experiments current topics in bio om a scientific point	gineering exper op their own pr basics of structu al, cellular and r n and membran n their physical n publications i ochemical analy	iments pint of view an ure-giving proc molecular biolo he biochemistry principles. n internationa ysis and related	esses ogy textbooks a y. I journals. d fields.	s well as the
- to independently prepare, present	and defend	d a scientific paper o	on a (given) curi	rent biochemio	cal-analytical to	pic.

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 03.04.2025



a)							
 Mechanisms of cellular signal transduction, signalling particular 	thways, receptors, genome						
 transcriptional regulation, epigenetics, stem cells 							
RNA structures, ribozyme, spliceosome, RNAses, riboswitches							
	Innate and adaptive immunity, haematopoiesis, phagocytosis, Toll-like receptors,						
 B- and T-cell receptors, cytokines, immunoglobulins, MH 							
 receptors, membrane domains, caveolae, ligand binding. 							
GPCR-associated diseases, heterotrimeric G proteins, sig							
 Second messengers (cAMP, cGMP, Ca2+, NO, inositol ph Brotain kinese families PKA, PKC, selmedulin, CoM kinese 							
 Protein kinase families, PKA, PKC, calmodulin, CaM kinas Receptor tyrosine kinases, growth factors, cytokine rece 							
 Ras family, MAP kinases, regulated proteolysis, secretase 							
Nucleolar receptors (steroid receptors, retinoid X recept							
Membrane transport, signal sequences, translocation to							
Protein modifications, unfolded protein response, secret							
Cytoskeleton (microtubules, actin, intermediate filament	ts), dynamics						
• Cell-cell, cell-matrix connections, extracellular matrix, ce	Il adhesion						
• Cell cycle and apoptosis: cyclins, CDKs, IAPs, Bcl proteins	, caspases, apoptosome, TNF, FasR						
 Neuronal signal transduction: basics in electrophysiology 	, ion channels						
b) and c)	$\langle O \rangle$						
Methods of protein expression							
 Principles and methods of protein isolation and identification 	ation						
Immune techniques in biochemistry							
 Spectroscopic methods in biochemistry 							
 Methods of protein structure analysis 							
Protein stability							
Protein dynamics							
 Chemical modification of proteins 							
 Biochemistry and biophysics of lipid membranes 							
Membrane proteins							
In vivo and in vitro studies of protein-protein and protein-lipid interactions							
Microscopic techniques							
Expression and protein characterisation in vivo	1						
Compulsory entrance requirements							
Recommended participation requirement(s) for the module and/or individual courses of the module							
Language(s) of instruction and examination	German or English						
Weight of the module grade in the overall grade	12/66; as elective: not graded						
Frequency of module offer	a) Only in the summer term b), c) Only in the winter term						
Reasons for compulsory attendance							
Person responsible for the module	UnivProf. Dr. Dirk Schneider						
	Bachelor of Science Molecular Biotechnology, Master of						
Transferability of the module to other degree programs	Science Chemistry, Master of Science Molecular Biotechnology						
Other	Note: The course(s) not chosen in the compulsory area, (a) or (b, c), can additionally be chosen in the elective area.						

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Module BCF-P	Practica	al Course in Mol	ecular Biolog	gy and	[Modul-	Kennnummer]
	Biochemistry					
Mandatory or elective Module	Elective				-	
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration (according to course plan)	1 Semest	er				
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints
a) Practical Course "Molecular Biology and Biochemistry"	APr	1 (1)	М	9	40,5 h	4,5
b) Supporting seminar to a)	S	1 (1)	М	1	34,5 h	1,5
In order to complete the module, yo	u have to	fulfil the following	requirements:			
Compulsory Attendance	APr					
Active participation	According	g to § 5 para. 3			0	
Coursework						
Module examination				.0		
Qualification Goals, learning outcom	ne, compe	tences				
 apply complex biocher to carry out experiment to work carefully and it interpret the results of apply effective time and to critically question a Contents a) Molecular biology: Promodified bacteria. Generation of a bacter Heterologous protein Characterisation of the 2D gel electrophoresis physical parameters (in Analysis of the phosphere Staining techniques for Purification of lysozymer check the function of the The student elaborate the presentation. The student analyses a 	nts indepenn n a coordi f their expond current to nd resource current to nd scientif oduction of rial strain s expression e protein a scientric protein g norylation protein g ne: ion excl the enzymous s and pres	ndently and on thei nated manner. eriments correctly a e management. opic in biochemistry ically discuss preser in an expression plas witable for protein of a nexpression plas witable for protein of a cill swith diffe point, size) for two- pattern of a stress p rels hange chromatogra e ents a given, curren	r own responsik and document th and defend it in nted seminar lea mid, production expression. in purification. rent stressors (H dimensional sep rotein by 2D ge phy, protein pre-	ility using cou hem in an app a a discussion i ctures. a and characte heat, oxidants, baration of con I electrophore ecipitation, SD emistry and fa	ropriate form. in front of the risation of gen , etc.), use of p mplex protein esis and Wester S-PAGE, photo ces the audien	entire audience. etically rotein-specific mixtures. rn blot analysis metric assay to ice to discuss
Compulsory entrance requirements						
Recommended participation require and/or individual courses of the mo		or the module				
Language(s) of instruction and exam	ination		German or English			
Weight of the module grade in the o	overall grad	de	Not graded			
Frequency of module offer			Every term in t	he lecture-fre	e period	
Reasons for compulsory attendance			According to HochSchG § 26 Para. 2 (7), Practical Course			
Reasons for compulsory attendance			According to H	ochSchG § 26	Para. 2 (7), Pra	actical Course



Transferability of the module to other degree programs	Bachelor of Science Molecular Biotechnology, Master of Science Chemistry
Other	Only in the lecture-free period

without Buarantee

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	Organic Chemistry [Modul-Kennnummer]					ennnummer]
Mandatory or elective Module	M					
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration						
(according to course plan)	1 Semest	er				
Courses/ Learning formats	Regular term when starting in Winter term (Summer term) Mandatory/ elective Contact Time (SWH) Self Study Creditpoint			Creditpoints		
a) Lecture "Aromatic/heterocyclic compounds"	L	1 (1)	м	2	69,0 h	3
b) Supporting exercise to a)	E	1 (1)	М	1	34,5 h	1,5
c) Supporting seminar (Trainee Seminar)	S	1 (1)	М	1	34,5	1,5
In order to complete the module, yo	u have to	fulfil the following	requirements:			•
Compulsory Attendance	s	·	-			
Active participation		to § 5 para. 3			0	
Coursework	According	, to 3 5 para. 5				
Module examination	Usually w and b)	ritten exam (120 m	in), alternatively	/ oral exam (30	0 min) on the co	ontents of a)
Qualification Goals, learning outcom	e. compet	tences				
 (aromatics/heterocycles) and acquires students are then able to reproduce in-depth kn concepts and methods to work out and deepee to establish connection chemistry within the suto transfer the learned to identify problems in work out possible solu evaluate them critically present their results in in discussions to critically question and c) The specialisation unit serves indiversearch. The students are able to, work independently or develop their preparat to analyse the results of expand their methodo 	owledge fi from these in content his and link ubject and lecture co the devel tions inder a compre a compre idual speci n research ive skills ir of indepen logical kno heir exper	rom the field of aro se fields and classify s from the field of a s between topics ar with related subject ontents to unknown opment of synthesi pendently by linking hensible manner ar e the solution strate ialisation and perso -related topics in pr ndependently dent literature rese	matics and hete them with rega iromatics and he nd contents fror ct areas. In tasks is strategies and g the acquired k and using scientif egies developed nal profile build reparative organ earch. enting new appa ement them ind	erocyclic chem ard to their sig eterocyclic che n the field of a in the answer nowledge with ically correct t I. ing in prepara nic chemistry, aratus and ana	istry, describe r nificance emistry indepen aromatics and h ring of complex n their own idea terminology and tion for later in	nodern Idently. eterocyclic questions, to as and to d defend them dependent

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a)

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α)							
 Aromaticity (criteria), systematic treatment of annulenes and properties of selected systems, 	s, non-alternating systems, PAHs, methods of preparation						
	Classification and nomenclatures of heterocycles, physical properties (solubility, pKs, dipole moments,)						
	roatoms, medium rings with up to four heteroatoms, seven-						
and eight-membered rings in their occurrence and produ							
 Application as active substances and in materials science 	2.						
b)							
Consolidation of the lecture material and applications in	transfer exercises						
c)							
 Preparative methods, reagents in organic synthesis, reac 	tion types and reaction mechanisms						
Compulsory entrance requirements							
Recommended participation requirement(s) for the module							
and/or individual courses of the module							
Language(s) of instruction and examination	German or English						
Weight of the module grade in the overall grade	12/66						
Frequency of module offer	Every term						
	Seminar according to § 5 para. 5: The learning objectives						
	are based on direct interaction between students. In						
Reasons for compulsory attendance	addition to practical professional competence, important						
	learning objectives are literature research, presentation						
	and leading discussions.						
Person responsible for the module	N.N. O						
Transferability of the module to other degree programs	Master of Science Chemistry						
	Recommended literature:						
	Gilchrist: Heterocyclenchemie,						
Other	Joule/Mills: Heterocyclic Chemistry,						
	Brückner: Reaktionsmechanismen						

hormation

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Module OCF-P	Practica	al Course on Mo	lecule Synth	esis	[Modul-I	Kennnummer]	
Mandatory or elective Module	м						
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	er					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
Practical Course "Molecule Synthesis"	APr	2 (2)	М	12	54 h	6	
In order to complete the module, yo	u have to	fulfil the following	requirements:				
Compulsory Attendance	APr				~	0	
Active participation	According	to § 5 para. 3					
Coursework							
Module examination					0		
Qualification Goals, learning outcom	e, compet	ences					
 develop their preparat analyse the results of i extend their methodol work out and plan thei debate with their supe work out solutions who assess the safety aspect develop their English la plan tasks together in a work responsibly in a t analyse and correct the assess and optimise th Contents Preparation of 3-4 research-related p from current chemical journals or Orget	ndepende ogical kno r experime rvisors hov en dealing cts of chen anguage sk a team and eam and t e experime e results o	nt literature researd wledge by impleme ents and to impleme w to carry out the e with scientific prob nicals and experime cills through English d to carry out prelin o handle hazardous ental results on the f the experiments.	enting new appa ent them indep xperiments and elems, nts and take ap -language litera hinary work, substances, basis of theoret	endently, I to correct the propriate acti ture and supe cical knowledg	em, on, ervisors, ge through tech	nical literature,	
Compulsory entrance requirements							
Recommended participation require and/or individual courses of the modest terms and the modest of th		or the module					
Language(s) of instruction and exam	ination		German or English				
Weight of the module grade in the o	verall grad	le	Not graded				
Frequency of module offer			Every term				
Reasons for compulsory attendance			According to H	ochSchG § 26	Para. 2 (7), Pra	ctical Course	
Person responsible for the module			UnivProf. Dr.	Till Opatz			
Transferability of the module to othe	er degree	programs	Master of Scier	nce Chemistry	1		
Other			JouleBrück	rist: Heterocy /Mills: Hetero (ner: Reaktion nic Synthesis,	clenchemie, ocyclic Chemisti Ismechanismer Organic Reacti	n	

Module PMC2	Pharma	acology for Natu	Iral Scientist	s	[Modul-H	(ennnummer]
Mandatory or elective Module	м				<u>_</u>	
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration (according to course plan)	1 Semest	ter				
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoint
a) Lecture "Pharmacology for Natural Scientists"	L	1 (1)	М	3	103,5 h	4,5
b) Supporting seminar to a)	S	1 (1)	М	1	34,5 h	1,5
In order to complete the module, yo	u have to	fulfil the following	requirements:			
Compulsory Attendance						
Active participation	According	g to § 5 para. 3				
Coursework					0	
Module examination	Usually w and b)	ritten exam (60 mir	n), alternatively	oral exam (15	min) on the co	ntents of a)
Qualification Goals, learning outcom	ne, compe	tences				
 to show connections be and molecular biology to name important dru adverse effects and im Contents a) Principles of pharmacodyna Principles of pharmacodyna Principles of pharmacodyna Drinciples of pharmacodyna Drug-drug interactions Important transmitters Drugs for the treatment or 	, ugs for the teractions) amics trics etics prevention	treatment of comn n of important disea	non diseases an	d their most ir	nportant prope	erties (includii
 cardiovascular diseases, dia b) processes of drug research Job profiles for natural scie 	, developn	nent and applicatior	١,			
Compulsory entrance requirements						
Recommended participation require and/or individual courses of the mo		or the module				
Language(s) of instruction and exam			German or Eng	lish		
Weight of the module grade in the o		de	6/66			
Frequency of module offer	Ū		Every term			
Reasons for compulsory attendance						
Person responsible for the module			UnivProf. Dr.	Andrea Pautz		
Transferability of the module to oth	er degree	programs				
Other	er wegree	F. 60. 0110	b) Can also be		he form of a bl	ock seminar

an online seminar.



Module PMC3	Pharma	aceutical Science	es for Natura	l Scientists	[Modul-I	(ennnummer]
Mandatory or elective Module	м				<u> </u>	
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration (according to course plan)	1 Semest	er				
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints
a) Lecture "Principles and Special Aspects of Drug Design"	L	2 (2)	М	2	69 h	3
b) Supporting seminar to a)	S	2 (2)	М	2	69 h	3
In order to complete the module, yo	ou have to	fulfil the following	requirements:			
Compulsory Attendance						
Active participation	According	g to § 5 para. 3				
Coursework					0	
Module examination	Usually w and b)	ritten exam (120 m	in), alternatively	y oral exam (3)	0 min) on the c	ontents of a)
Qualification Goals, learning outcon	ne, compe	tences				
 discuss medicinal chere to assess and classify in Contents a) and b) Principles of drug desi Drug targets Hit-to-lead development Optimisation of lead s Concepts of structure Analysis of target-ligat ADME-tox properties PK-PD optimisation of Development of coval Determination and ca Drug monitoring Literature / publicatio Computer-aided drug pharmacophore mode structure optimisation 	ign ent tructures -activity re nd interact of drugs; ro drugs (pha ent drugs lculation or ns on curre design (CA elling, proto n, generatio	aches to drug devel lationships and SAR ions at molecular at elationship with che armacodynamics-ph f physicochemical p ent topics in drug de DD): e.g. visualisati ein-ligand docking, v on of 3D structures	analysis nd atomic level mical structure armacokinetics arameters of ac evelopment on and analysis virtual screening of small molecu) tive substance of protein-liga g, scoring, QSA les, force field	nd complexes, R, ADME mode s, MD simulatio	elling, lead ons,
design, target assessmBiologics and antibodi	ies	argets				
Nucleic acids as drugs Drug delivery, drug ta Combinatorial approa Compulsory entrance requirements	rgeting, dr ches	ug transport				
 Nucleic acids as drugs Drug delivery, drug ta Combinatorial approa 	rgeting, dr ches ement(s) fo					

Weight of the module grade in the overall grade	6/66
Frequency of module offer	Every term
Reasons for compulsory attendance	
Person responsible for the module	UnivProf. Dr. Tanja Schirmeister
Transferability of the module to other degree programs	
Other	
hormation	noutebarante



Module FMP	Researc	ch Project			[Modul-ł	(ennnummer]
Mandatory or elective Module	м					
Creditpoints (LP) and workload	12 LP = 3	60 h				
Module duration (according to course plan)	1 Semest	Semester				
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints
a) Practical Course "Research Paper"	Apr	3 (3)	М	22	99,0 h	11
b) Supporting Seminar "Guidance for independent scientific work"	S	3 (3)	М	1	19,5 h	
In order to complete the module, yo	u have to	fulfil the following	requirements:			
Compulsory Attendance	Apr (acco	rding to the task an	d agreement wi	th the superv	isor)	
Active participation	According	According to § 5 para. 3, presentation on the research module (30 min)				
Coursework	Maintena	Maintenance of a laboratory notebook				
Module examination	Research	report				
Qualification Goals, learning outcom	e, compe	tences				
experiments under supervision. They recorded reproducibly in a laboratory By working in a working group, the st Contents a) Participation in a current research	book and udents ex	interpreted in the pand their commun	final report, tak ication and tear	ing into accou mwork skills.	nt current rese	
 b) Introduction to planning, execution in a report (protocol) and seminar lec 		umentation of more	complex scient	ific experimer	nts. Presentatio	n of the results
Compulsory entrance requirements			According to P	0		
Recommended participation require and/or individual courses of the modest the modest set the set of the modest set the set of	• •	or the module				
Language(s) of instruction and exam	ination		German or English			
Weight of the module grade in the o	verall grad	de	Not graded			
Frequency of module offer			Every term			
Reasons for compulsory attendance research			According to HochSchG § 26 Abs. 2 (7), scientific (practical) research work/internship (according to assignment and agreement with the supervisor).			
Person responsible for the module			All full-time lec	turers involve	d in the degree	e programme
Transferability of the module to othe	er degree	programs	Master of Scier	nce Biomedici	nal Chemistry	
Other			The module tal Individual supe in a participatir to semester. Ex application.	ervision; the ning working gro	umber of interr oup may vary fr	om semester



Module MSC	Master	Thesis			[Modul-I	(ennnummer]	
Mandatory or elective Module	м						
Creditpoints (LP) and workload	30 LP = 9	00 h					
Module duration (according to course plan)	1 Semest	er					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
Master Thesis		4 (4)	М	6 months all day	900 h	30	
In order to complete the module, yo	u have to	fulfil the following	requirements:			•	
Compulsory Attendance	Master th	esis (according to a	ssignment and	agreement wit	th the supervis	or)	
Active participation	According	to § 5 Para. 3, pres	entation on the	Master's the	sis (30 min)		
Coursework	Maintena	nce of a laboratory	notebook				
Module examination	Master th	esis			0		
Qualification Goals, learning outcom	e, compet	ences					
them in the light of the relevant litera answering questions on the topic as y Contents				5			
Master's thesis: Composition of a scie introduction including objectives, ma	terial & m	ethods as well as re	sults, discussion	n, bibliography	; an appendix		
to document further primary data. Pr	resentation				ssion.		
Compulsory entrance requirements			According to §	15 para. 4			
Recommended participation require and/or individual courses of the modest the modest set the set of the modest set the set of		or the module					
Language(s) of instruction and exam	ination		German or Eng	lish			
Weight of the module grade in the o	verall grad	le	30/66				
Frequency of module offer			Every term				
Reasons for compulsory attendance			According to HochSchG § 26 Para. 2 (7), scientific (practical) research work/internship (according to assignment and agreement with the supervisor).				
Person responsible for the module			All full-time lecturers involved in the degree programme				
Transferability of the module to othe	er degree	programs	Master of Scier	nce Biomedicir	nal Chemistry		
Other			The module tal Individual supe in a participatin to semester. Ex application.	ervision; the nung working gro	umber of interr oup may vary fr	nships offered fom semester	

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Elective Area

All elective modules do not count towards the final grade.

Module NC	Chemis	try of Natural P	roducts		[Modul-k	(ennnummer]
Mandatory or elective Module	Elective				<u> </u>	
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration (according to course plan)	1 Semest	er				
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints
a) Lecture "Chemistry of Natural Products"	L	2 (1)	М	2	69 h	3
b) Supporting exercise to a)	E	2 (1)	М	1	34,5 h	1,5
c) Seminar "Retrosynthesis"	S	2 (1)	М	1	34,5 h	1,5
In order to complete the module, yo	u have to	fulfil the following	requirements:			•
Compulsory Attendance)	
Active participation	According	to § 5 para. 3		Š V		
Coursework			(5		
Module examination	Usually w b) and c)	ritten exam (120 m	in), alternativel	y oral exam (3	0 min) on the c	ontents of a),
Qualification Goals, learning outcom	e, compet	ences				
 to establish connection within the subject and to transfer the content to identify problems in independently work of evaluate them 	with relat s of the le the devel at possible	ed subject areas. cture to unknown t opment of synthesi solutions by linking	asks. s strategies and g the acquired k	l in the answei nowledge witl	ring of complex	questions, to
to critically question ar Contents	nd evaluat	e the solution strate	egies developed	1.		
 a) Organic Chemistry 5: Classe Amino acids, peptides and p Terpenes and steroids Lipids and eicosanoids Polyketides carbohydrates Biogenic amines and alkaloi nitrogenous cofactors of pr synthesis and biosynthesis ab) consolidation of the lecture matic) 	oroteins, p ds oteins and analys	eptide synthesis is of natural produc	rts.	es and nucleic	: acids, nucleic a	acid synthesis.
Methods of organic synthesis an	d retrosyr	thesis on concrete	examples.			
Compulsory entrance requirements						
Recommended participation require and/or individual courses of the mod		or the module				

Language(s) of instruction and examination	German or English
Weight of the module grade in the overall grade	Not graded
Frequency of module offer	Only in the summer term
Reasons for compulsory attendance	
Person responsible for the module	UnivProf. Dr. Till Opatz
Transferability of the module to other degree programs	Master of Science Chemistry
Other	Recommended Literature: Nuhn: Naturstoffchemie Habermehl/Hammann/Krebs/Ternes: Naturstoffchemie
hormation	outerante

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Module RPC	Radiopharmaceutical Chemistry [Modul-Kennr				ennnummer]	
Mandatory or elective Module	Elective					
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration (according to course plan)	2 Semest	er				
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints
a) Lecture "Radiopharmaceutical Chemistry 1"	L	1 (2)	М	2	69h	3
b) Lecture "Radiopharmaceutical Chemistry 2"	L	2 (1)	м	2	69 h	3
In order to complete the module, y	ou have to	fulfil the following	requirements:			
Compulsory Attendance						
Active participation	According	g to § 5 para. 3				
Coursework						
Module examination	Usually or and b)	ral exam (30 min), a	lternatively wri	tten exam (12	0 min) on the co	ontents of a)
Qualification Goals, learning outcor	ne, compe	tences				
 preclinical a radionuclida radiopharm properties, 	I Chemistry n and basic nd clinical e productio naceutical p production logy, neuro	(RPC) are offered a s of RPC: decay mod imaging techniques n in RPC: cyclotron, rocedures in diagno , labelling chemistry ology and other field	of new radiopha s block courses des, shielding & , reactor & gene ostics and therap y & application ds of application	over 2 semest detection. rator, py: SPECT, PET of relevant num	ers. Contents a	re:
				chemistry .		
Compulsory entrance requirements Recommended participation requir and/or individual courses of the mo	ement(s) fo	or the module	Module "Introd	duction to Nuc	lear Chemistry	,
Language(s) of instruction and exam	nination		German or English			
Weight of the module grade in the	overall grad	de	Not graded			
Frequency of module offer			a) Only in the winter term b) Only in the summer term			
Reasons for compulsory attendance	9					
Person responsible for the module			UnivProf. Pat	rick Riß		
Transferability of the module to oth	ner degree	programs	Master of Scier	nce Chemistry		
Other						

Module MCP	Dractics	Course Salast	od Asposts o	fMadicipal	[Modul-	(ennnummer]		
		Practical Course Selected Aspects of Medicinal [Modul-Kennnummer] Chemistry						
Mandatory or elective Module	Elective	- /						
Creditpoints (LP) and workload	6 LP = 18	0 h						
Module duration	01.10	• · · ·						
(according to course plan)	1 Semest		[[
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints		
a) Practical Course "Selected Aspects of Medicinal Chemistry"	APr	1 - 3 (1 - 3)	М	6	117 h	6		
In order to complete the module, yo	u have to	fulfil the following	requirements:					
Compulsory Attendance	APr							
Active participation	According	to § 5 para. 3						
Coursework								
Module examination								
Qualification Goals, learning outcom	e, compet	ences						
 use the subject-specific point out connections optimisation of active set to work out and presending discuss medicinal chemendiate to evaluate and classifice to understand and apply understand and apply and and apply and and apply and and apply and analyse exists 	and differd substances nit a medic nistry topid y new app ly method experimer methods f methods f methods f	ences between diffe inal chemistry topic cs appropriately roaches to drug dev s of drug and drug atal and theoretical or biotransformatic or determining the or computer-aided al data	erent approache independently velopment and o analysis methods for the on of active subs efficacy of biolo drug design.	optimisation e determinatio tances and fo gically active	on of physico-cl r drug monitor substances.	hemical drug		
Contents								
The practical course includes the folic Visulisation and analys Calculation of physicoo Pharmacophore mode Protein-ligand docking Homology modelling Lead structure optimis Biotransformation and Stability studies of drug Drug and drug substan Determination of phys Drug monitoring Enzyme kinetics, ligand Quantitative HPLC for o	is of prote chemical a ls ation determin gs ce analysis icochemic d binding s	in-ligand complexe nd pharmacokinetio ation of metabolite al and pharmacokin tudies	s parameters	5				
Compulsory entrance requirements								
Recommended participation require and/or individual courses of the mod		or the module						
Language(s) of instruction and exam	ination		German or Eng	lish				
Weight of the module grade in the o	verall grad	le	Not graded					

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Frequency of module offer	Every term in the lecture-free period
Reasons for compulsory attendance	According to HochSchG § 26 Para. 2 (7), Practical Course
Person responsible for the module	UnivProf. Dr. Tanja Schirmeister
Transferability of the module to other degree programs	
Other	Block internship (3 weeks full time) during the lecture-free period

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Module BAC	Bioinorganic Chemistry [Modul-Kennnum						
Mandatory or elective Module	Elective				<u> </u>		
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	er					
Courses/ Learning formats	TypeRegular term when starting in Winter term (Summer term)Mandatory/ electiveContact Time (SWH)Self Study						
a) Lecture "Bioinorganic Chemistry"	L	2 (1 o. 3)	М	3	103 <i>,</i> 5 h	4,5	
b) Supporting Seminar to a)	S	2 (1 o. 3)	М	1	34,5 h	1,5	
In order to complete the module, yo	u have to	fulfil the following	requirements:				
Compulsory Attendance	S				5		
Active participation	According	to § 5 para. 3					
Coursework					0		
Module examination	Usually w	ritten exam (120 m	in), alternatively	y oral exam (3	0 min) on the c	ontents of a)	
Qualification Goals, learning outcom	ne, compe	ences					
 can establish connecti disciplines, have gained an unders Contents Bioinorganic chemistry is a cross-sect The lecture serves to identify the spet Biological processes such as photosyness.	tional disci cific roles nthesis or	the significance of pline of biochemist of certain metal ion cellular respiration	metal ions in liv ry and coordina s in chemical-bi are discussed.	tion chemistry	r. Icesses.		
Selected examples of metalloprotein in more detail as well as electron tran		-			•	are discusse	
Compulsory entrance requirements							
Recommended participation require and/or individual courses of the mo		or the module					
Language(s) of instruction and exam	ination		German or Eng	lish			
Weight of the module grade in the o	verall grad	de	Not graded				
Frequency of module offer			Only in the sun	nmer term			
Reasons for compulsory attendance Upper seminar according to § 5 para. 5: The objectives are based on direct interaction students. In addition to practical profession important learning objectives are literature presentation and leading discussions.					t interaction be ical profession s are literature	etween al competence	
Person responsible for the module			UnivProf. Dr.	Eva Rentschle	r		
Transferability of the module to oth	er degree	programs	Master of Scier	nce Chemistry			
Other							



Module BPC	Biophy	sical Chemistry			[Modul-k	(ennnummer]		
Mandatory or elective Module	Elective							
Creditpoints (LP) and workload	6 LP = 18	0 h						
Module duration (according to course plan)	1 Semest	L Semester						
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints		
a) Lecture "Biophysical Chemistry"	L	2 (1)	М	2	69 h	3		
b) Supporting exercise to a)	E	2 (1)	М	2	69 h	3		
In order to complete the module, yo	u have to	fulfil the following	requirements:					
Compulsory Attendance					X	0		
Active participation	According	g to § 5 para. 3						
Coursework								
Module examination	Usually w	ritten exam (120 mi	in), alternatively	y oral exam (3	0 min)			
Qualification Goals, learning outcom	ne, compet	tences						
fields. The students should be able to get to the bottom of unknown pheno		e appropriate metho	ods for new exp	erimental que	stions in order	to successfully		
Contents								
 a) Basics of modern biophysical meth Membrane transport, phase transitio motors, single molecule techniques, the drug discovery process. b) In-depth or supplementary topics 	ons in mem Raman sca	branes, nanopartic ttering, thermodyn	le sensors, rate amics of chemic	equations and cal bonds, physical bonds, physi	l dynamics in ce sical-chemical p			
Compulsory entrance requirements								
Recommended participation require and/or individual courses of the mod		or the module						
Language(s) of instruction and exam	ination		German or English					
Weight of the module grade in the o	verall grad	de	Not graded					
Frequency of module offer			Only in the sun	nmer term				
Reasons for compulsory attendance								
Person responsible for the module			UnivProf. Dr.	Carsten Sönni	chsen			
Transferability of the module to oth	er degree	programs	Master of Science Chemistry					
Other								
11.			1					

Module Tox1	Toxicol	ogy 1			I	[Modul-K	(ennnummer]
Mandatory or elective Module	Elective	Elective					
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	er					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self	Study	Creditpoints
a) Lecture "General Toxicology"	L	1 - 3 (1 - 3)	М	2	69	9 h	3
b) Seminar "Molecular and Cellular Toxicology"	S	1 - 3 (1 - 3)	М	2	69	9 h	3
In order to complete the module, yo	ou have to	fulfil the following	requirements:				
Compulsory Attendance	S						
Active participation	According	g to § 5 para. 3					
Coursework					0		
Module examination	Usually w	ritten exam (60 mir	ı), alternatively	oral exam (15	min) or	n the cor	ntents of a)
Qualification Goals, learning outcom	ne, compet	tences					
 name the treated biog describe the symptom apply their knowledge toxin and (theoreticall apply their knowledge fine dust, carcinogenic b) The students are able to independently process original publications of Contents a) Basics of toxicology Toxic effects (receptor Chemical mutagenesis Heavy metals, solvent Biological toxins (plant Biocides, toxic drugs, therapy of poisoning b) In the seminar, the topics of chemit the field of genotoxicology are discuss cytotoxicity, mechanisms of carcinoged 	atology of to conclud y) to carry to be able c substance s original p rally, evalu site theor s and carcin s, alcohols, t, bacterial coxicology ical mutag ssed. This i	the different poison de on the basis of the out an appropriate to assess current the es, new psychoactiv ublications from the late them and discu y, systemic toxins, of nogenesis, genotoxi , food toxins, anima of drug consumptio enesis and carcinog ncludes topics such	nings. ne symptomatol therapy. opics in the med e substances wh e field of molect ss them critical concentration to ns l toxins) n, enesis are conti as ageing resea	ogy a specific dia. This incluc nich are releva ular toxicology y within the g oxins, summat	poisoni des topi ant to th y, prese roup. tion toxi	ing and t cs such a ne public nt the da ins etc) ginal pul	the triggering as exposure to c. ata of the blications from
Compulsory entrance requirements							
Recommended participation require and/or individual courses of the mo		or the module					
Language(s) of instruction and exam	ination		German				
Weight of the module grade in the o	overall grad	de	Not graded				
Frequency of module offer			Every term				
Reasons for compulsory attendance			Seminar accord are based on d addition to pra learning object	irect interactio ctical professi ives are litera	on betw onal co	veen stu mpetend	dents. In ce, important

and leading discussions.

Univ.-Prof. Dr. Markus Christmann

Person responsible for the module

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Transferability of the module to other degree programs	Bachelor of Science Biomedical Chemistry
Other	

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Module ToxP	Toxicol	ogy 2			[Modul-k	(ennnummer]
Mandatory or elective Module	Elective				<u> </u>	
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration (according to course plan)	1 Semest	er				
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints
Practical Course "Molecular methods in toxicology"	APr	2 (1 o. 3)	М	6	117 h	6
In order to complete the module, yo	u have to	fulfil the following	requirements:			
Compulsory Attendance	APr					0
Active participation	According	g to § 5 para. 3				
Coursework						
Module examination	Written e	xam (30min)			0	
Qualification Goals, learning outcom	e, compe	tences				
 to carry out toxicologic analyses, microscopic e to adequately record a Contents In the context of the practical course, mutagenesis, genotoxicity, DNA repair relevant techniques. Mechanisms of cytotox Investigation of genotox Toxicity and mutagenic Transcriptional and epir Principles of toxicologic molecular causes of ag post-translational mod 	examination nd evalua students r and cell kicity (apo oxic effect: city assays genetic re cal risk ass eing	ons). te the performance should acquire furtl death mechanisms, ptosis, necrosis, aut s: SCE, aberration, p , Ames test gulatory mechanism sessment	and results of p her theoretical l as well as pract cophagy,) point mutation a	knowledge ab	tigations. out mechanism	s of
Compulsory entrance requirements		i proteins	Module Toxico	logv 1		
Recommended participation require and/or individual courses of the mod		or the module				
Language(s) of instruction and exami	ination		German			
Weight of the module grade in the o	verall grad	de	Not graded			
Frequency of module offer			Only in the sun	nmer term		
Reasons for compulsory attendance			According to H	ochSchG § 26	Para. 2 (7), Pra	ctical Course
Person responsible for the module			UnivProf. Dr.	Markus Christ	mann	
Transferability of the module to othe	er degree	programs	Bachelor of Sci	ence Biomedi	cal Chemistry	
Other					· ·	

Module Immun1	Immun	ological Principl	<mark>es</mark>		[Modul-K	ennnummer]	
Mandatory or elective Module	Elective	Elective					
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	er					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Lecture "Immunological Principles"	L	2 (1 o. 3)	М	2	69 h	3	
b) Supporting seminar to a)	S	2 (1 o. 3)	М	2	69 h	3	
In order to complete the module, yo	u have to	fulfil the following	requirements:				
Compulsory Attendance							
Active participation	According	to § 5 para. 3					
Coursework					0		
Module examination	Usually w and b)	ritten exam (90 mir	n), alternatively	oral exam (30	min) on the co	ntents of a)	
Qualification Goals, learning outcom	e, compet	ences					
 establish the significan understand the special allergies and autoimme independently develop discuss immunological Contents Lecture and seminar include the follo Organs and cells of the Mechanisms of innate Development and funct Development and funct Tolerance mechanisms Importance of the maj Genetic models in imm Signal transduction in I Mucosal immune system 	significan une diseas and prese topics app wing topic immune immunity tion of B-o tion of T c in histoco nunology ymphocyt	ce of immunologica es. ent a (given) immur propriately. ss: system; Haematopo cells and antibodies ells mpatibility complex es	il research for the	ne developme			
Compulsory entrance requirements							
Recommended participation require and/or individual courses of the mod		or the module					
Language(s) of instruction and exam	ination		German				
Weight of the module grade in the o	verall grad	de	Not graded				
Frequency of module offer			Only in the sun	nmer term			
Reasons for compulsory attendance							
Person responsible for the module			apl. Prof. Dr. M	lichael Stasser	1		
Transferability of the module to othe	er degree	programs			cal Chemistry, N cience Biomedic		
Other			-		st half of the se the second half	mester,	

Module Immun2	Practical Exercises in Immunology [Modul-Ken					Kennnummer]	
Mandatory or elective Module	Elective		<u>_</u>				
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semester						
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Practical Exercise "Immunology"	E	2 (1 o. 3)	М	8	96 h	6	
In order to complete the module, yo	u have to	fulfil the following	requirements:				
Compulsory Attendance	E					20	
Active participation	According	to § 5 para. 3			X \	0	
Coursework					5		
Module examination							
Qualification Goals, learning outcom	e, compet	ences					
 document the results of statistical methods agree on individual wo reproduce and explain Contents In the exercises, the following content Quantification of cytok Detection of mediator Determination of the a Identification and enrice Blood group serology Enrichment and activation 	rk steps, p the theor ts will be v cines by EL release fro activity of r chment of	lan them together a y on which the expension worked on experime ISA and qRT-PCR om activated mast of reporter genes defined cell popula	and implement eriments are bas entally: sells	them in a coordinate		-	
Compulsory entrance requirements			Module "Immu	unological Prin	ciples"		
Recommended participation require and/or individual courses of the mod		or the module					
Language(s) of instruction and exam	ination		German				
Weight of the module grade in the o	verall grad	le	Not graded				
Frequency of module offer			Only in the sun	nmer term dur	ing the lecture	-free period	
Reasons for compulsory attendance			According to H	ochSchG § 26	Abs. 2 (7), prac	tical exercise	
Person responsible for the module			apl. Pof. Dr. Mi	ichael Stassen			
Transferability of the module to othe	er degree	programs	Bachelor of Science Biomedical Chemistry, Master of Science Biology, Master of Science Biomedicine				
Other			Registration re course during t			er; block	

Module PB1	Pharma	aceutical Biology	[Modul-k	[Modul-Kennnummer]				
Mandatory or elective Module	Elective	Elective						
Creditpoints (LP) and workload	6 LP = 18	0 h						
Module duration (according to course plan)	1 Semest	L Semester						
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints		
a) Lecture "Pharmaceutical Biology I, II or III"	L	1 - 3 (1 - 3)	м	2	69 h	3		
b) Seminar "Biogenic Medicinal Products (Antibiotics, Genetically Engineered Medicinal Products)"	S	1 - 3 (1 - 3)	М	2	69 h	3		
In order to complete the module, yo	u have to	fulfil the following	requirements:					
Compulsory Attendance	S			5	0			
Active participation	According	g to § 5 para. 3						
Coursework								
Module examination	Usually w and b)	ritten exam (120 m	in), alternativel	y oral exam (3	0 min) on the c	ontents of a)		
Qualification Goals, learning outcom	ie, compe	tences						
a) and b) The students are able to, classify and reproduce	basic the	pretical knowledge	of pharmaceutio	cal biology				
Contents								
 a) Medicinal plants, biogenic and nor b) Antibiotics, plant cytostatics, gene pharmaceutical biology 	-				abolites, techn	ical methods of		
Compulsory entrance requirements								
Recommended participation require and/or individual courses of the mo		or the module						
Language(s) of instruction and exam	ination		German					
Weight of the module grade in the o	verall grad	de	Not graded					
Frequency of module offer			a) Every semester (I, II and III alternately) b) Every semester					
Reasons for compulsory attendance			Seminar according to § 5 para. 5: The learning objectives are based on direct interaction between students. In addition to practical professional competence, important learning objectives are literature research, presentation and leading discussions.					
Person responsible for the module			UnivProf. Dr. Thomas Efferth					
Transferability of the module to oth	er degree	programs	Bachelor of Science Biomedical Chemistry					
Other								

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Module PBP	Practical Course in Pharamaceutical Biology [Modul-Kenn							
Mandatory or elective Module	Elective							
Creditpoints (LP) and workload	6 LP = 18	0 h						
Module duration (according to course plan)	1 Semest	ter						
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints		
a) Practical course "Pharmaceutical Biology III: Biological and Phytochemical Investigations".	APr	1 - 3 (1 - 3)	М	6	117 h	6		
In order to complete the module, yo	ou have to	fulfil the following	requirements:					
Compulsory Attendance	APr							
Active participation	According	g to § 5 para. 3						
Coursework								
Module examination								
Qualification Goals, learning outcon	ne, compe	tences						
apply basic molecular Contents Biological and phytochemical studies MS, HPLC, isolation of genomic DNA,	of medici	nal plants, identifica	tion of herbal c	drugs according	g to the pharm	acopoeia (DC),		
Compulsory entrance requirements			Module "Pharr	naceutical Bio	logy"			
Recommended participation require and/or individual courses of the mo		or the module						
Language(s) of instruction and exam	ination		German					
Weight of the module grade in the o	overall grad	de	Not graded					
Frequency of module offer			Every term					
Reasons for compulsory attendance			According to HochSchG § 26 Para. 2 (7), Practical Course					
Person responsible for the module			UnivProf. Dr. Thomas Efferth					
Transferability of the module to oth	er degree	programs	Bachelor of Sci	ence Biomedio	cal Chemistry			
Other								
heory								

Module MiBiT	Microb	Microbiology and Biotechnology					
Mandatory or elective Module	Elective				<u>_</u>		
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration	2 Semest	ters					
(according to course plan) Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Lecture "Microbiology"	L	1 o. 3 (2)	М	2	69 h	3	
b) Lecture "Biotechnology"	L	1 o. 3 (2)	М	2	69 h	3	
In order to complete the module, y	ou have to	fulfil the following	requirements:				
Compulsory Attendance					X	0	
Active participation	According	g to § 5 para. 3					
Coursework							
Module examination	b) As a ru Both exai of both e	v written exam (60 n le, lecture (25 min), minations must be p xaminations.	alternatively or	ral examinatio	n (30 min).	ithmetic mean	
Qualification Goals, learning outcome a) The students are able to	me, compe	tences					
 to name the most im to evaluate the import b) The students are able to apply in-depth knowl fermentation, proces Interpret biotechnologie extract scientific data to plan sophisticated to confidently assess Contents a) Microbiology:	rtance of ba edge in imp sing of prot ogical facts. I from datal biochemica	acteria in nature and portant sub-areas of ceins and secondary bases al and biotechnologi	for humans biotechnology metabolites fro cal experiments	(isolation and om submerged	cultures of fun		
 Structure of a bacteri Identification and cul Detection of mutatio Regulation in bacteria b) Biotechnology: Theory on the handling of microorg media optimisation for Isolation of biological Isolation of enzymes Characterisation of an 	ture techni ns; metabo a; structure anisms and or fermenta lly active ing from cultur	ques of bacteria lic physiology of bac and properties of b fermentation of mi- ations of microorgar gredients es of higher fungi	acteriophages croorganisms,				
Compulsory entrance requirements	-						
Recommended participation requir and/or individual courses of the me	ement(s) f	or the module					
Language(s) of instruction and exar	nination		German				
Weight of the module grade in the	overall gra	de	Not graded				
			a) Only in the v	vinter term			
Frequency of module offer			b) Only in the s				

Person responsible for the module	UnivProf. Dr. Ralf Heermann
Transferability of the module to other degree programs	Bachelor of Science Biomedical Chemistry
Other	

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Module TPhys	<mark>Animal</mark>	Physiology Physiology	[Modul-Kennnummer]			
Mandatory or elective Module	Elective				÷	
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration (according to course plan)	1 Semest	er				
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints
a) Lecture "Physiology, Neurobiology and Behaviour of Animals"	L	1 - 3 (1 - 3)	М	4	138 h	6
In order to complete the module, you	u have to	fulfil the following	requirements:			
Compulsory Attendance						0
Active participation	According	to § 5 para. 3				
Coursework						
Module examination	Usually w	ritten exam (60 mir	n), alternatively	oral exam (30	min)	
Qualification Goals, learning outcom	e, compet	ences				
 to express themselves Contents Function and interaction ecophysiological adapt Regulation of homeost Biochemistry of enzym Function and mode of a Cellular excitability, exc Sensory physiology (e.g Neurophysiology, learn Behavioural physiology Processes in muscle co 	on of organ ations to o asis es action of h citation pr g. sight, he ning and m r, orientat	ns extreme habitats normones ocesses, neuronal p earing, sense of bala nemory ion services, interna	processing mech ance, taste, sme al clock	nanisms II)		
Performance physiolog Computery entrance requirements	sy.					
Compulsory entrance requirements Recommended participation require and/or individual courses of the mod		or the module				
Language(s) of instruction and examination German						
Weight of the module grade in the or		le	Not graded			
Frequency of module offer			Every term			
Reasons for compulsory attendance						
Person responsible for the module			UnivProf. Dr.	Roland Strauß		
Transferability of the module to othe	er degree	programs	Bachelor of Sci Science Moleci			Bachelor of
Other						

Module PPhys	Plant Physiology				Plant Physiology [Modul-Kennnumm			(ennnummer]
Mandatory or elective Module	Elective		<u>.</u>					
Creditpoints (LP) and workload	6 LP = 18	0 h						
Module duration (according to course plan)	1 Semest	er						
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints		
a) Lecture "Plant Physiology"	L	1 - 3 (1 - 3)	М	4	138 h	6		
In order to complete the module, yo	u have to	fulfil the following	requirements:			•		
Compulsory Attendance						2		
Active participation	According	g to § 5 para. 3			X	0		
Coursework								
Module examination	Usually w	ritten exam (60 mir	n), alternatively	oral exam (30	min)			
Qualification Goals, learning outcom	ne, compe	tences						
 Functions of the comp primary and secondary photosynthetic and dis formation, transport, s Uptake and transport Metabolic cycles (espe Structure and function Regulation of plant de Light receptors, photo 	y reactions ssimilatory storage an of mineral ecially nitro of enzym velopment	of photosynthesis; energy metabolism d mobilisation of as s ogen cycle) es c, hormones, seed g	similates; lipid, ermination; pla	protein and ca nt cancer	irbohydrate me	etabolism;		
Water balance, water								
Compulsory entrance requirements								
Recommended participation require and/or individual courses of the mo		or the module						
Language(s) of instruction and exam	ination		German					
Weight of the module grade in the overall grade			Not graded					
Frequency of module offer			Every term					
Reasons for compulsory attendance								
Person responsible for the module			UnivProf. Dr.	Andreas Wach	nter			
Transferability of the module to oth	er degree	programs	Bachelor of Sci Science Molect			Bachelor of		
Other								



Module EC	Electro	chemistry	[Modul-Kennnummer]			
Mandatory or elective Module	Elective				<u>.</u>	
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration (according to course plan)	1 Semest	er				
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints
a) Lecture "Electrochemistry"	L	2 (1)	М	4	138 h	6
In order to complete the module, yo	u have to	fulfil the following	requirements:			•
Compulsory Attendance						2
Active participation	According	to § 5 para. 3			X '	0
Coursework					5	
Module examination	Usually w	ritten exam (120 mi	in), alternatively	v oral exam (3	0 min)	
Qualification Goals, learning outcom	e, compet	ences				
 have developed an aw interdisciplinary field. Contents Physical basics and ter potentials and current Electrode materials, el spectroelectrochemist Corrosion, electrocher Production of basic inco Cathode reactions (me Anode reactions (coup Natural product synthe Technical electroorgar Electrochemical surface 	ms (condu s). ectrolyte s nical millin organic che ediated sys lings, fluor esis nic synthes e treatme	ctivity in ionic syste cience, mediators, s theory g and machining; el emicals tems, direct metho rination, modern co is nt	ems; potentials a separators and lectroplating/m ds, technical ap	and structures cell geometrie etal deposition	at phase boun s; cyclic voltam	daries;
 Electropolymerisation, Ion exchangers Bioelectrochemistry, e 						
Compulsory entrance requirements						
Recommended participation require and/or individual courses of the mod		or the module				
Language(s) of instruction and examination German or English						
Weight of the module grade in the o	verall grad	de	Not graded			
Frequency of module offer			Only in the sun	nmer term		
Reasons for compulsory attendance						
Person responsible for the module			N.N.			
Transferability of the module to othe	er degree	programs	Master of Scier	nce Chemistry		
Other						

Module APP	Integra	ted Analytical-P	reparative L	ab Course	[Modul-K	(ennnummer]	
Mandatory or elective Module	Elective						
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	ter					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Lecture "Analytical Methods"	L	1 o. 2 (1 o. 2)	М	1	34,5 h	1,5	
b) Analytical Preparative Lab Course	APr	1 o. 2 (1 o. 2)	М	9	40,5 h	4,5	
In order to complete the module, yo	u have to	fulfil the following	requirements:	•			
Compulsory Attendance	APr						
Active participation	According	g to § 5 para. 3					
Coursework					0		
Module examination							
Qualification Goals, learning outcom	e comne	tences					
 work independently or evaluate the results of work out and plan theil debate with their super correct them, work out solutions wh assess the safety aspect develop their English late work responsibly in a t analyse and correct ex assess and optimise th 	their wor ir experim ervisors the en dealing cts of cher anguage s ream and t perimenta	k by analytical meth ents and implement e performance of th with scientific prob nicals and experime kills through English to handle hazardous al results based on th	ods. t them independ e experiments a lems and comb nts and take ap -language litera s substances, heoretical know	dently, and the analyt ine practice a propriate acti ture and supe	ical techniques nd theory, on, rvisors,	used and to	
Preparation of 2-4 research-related p e.g. from current chemical journals o with the analytical methods presente Depending on the preparation, labell	r Organic s d in the b	Syntheses. The obta lock lecture, among	ined pure subst others, and the	ances or subs	tance mixtures	are analysed	
Compulsory entrance requirements							
Recommended participation require and/or individual courses of the mod		or the module	Module "Practi	cal Course on	Molecular Synt	thesis"	
Language(s) of instruction and exam		German or English					
Weight of the module grade in the o	verall gra	de	Not graded				
Frequency of module offer Every term							
			- ,				
Reasons for compulsory attendance			According to H	ochSchG § 26	Para. 2 (7), Pra	ctical Course	

Master of Science Chemistry Recommended Literature:

Organic Syntheses, Organic Reactions, Houben-Weyl

Other

Transferability of the module to other degree programs

Module EM	Electro	ns in Molecules	[Modul-I	(ennnummer]			
Mandatory or elective Module	Elective						
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	er					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Lecture "Electrons in Molecules"	L	1 (2)	М	3	103,5 h	4,5	
b) Supporting exercise to a)	E	1 (2)	М	1	34,5 h	1,5	
In order to complete the module, yo	u have to	fulfil the following	requirements:				
Compulsory Attendance					X,	0	
Active participation	According	g to § 5 para. 3			5		
Coursework							
Module examination	Usually w	ritten exam (120 m	in), alternatively	y oral exam (3	0 min)		
Qualification Goals, learning outcom	ne, compe	tences					
 can establish connecti disciplines, have gained an unders sciences. 							
Contents							
Magnetic properties of organic mole concepts, application examples from properties of molecular compounds: from bio-inorganic chemistry. Introdu	bio-inorga Electron ti	anic chemistry, spin ransfer in discrete a	crossover comp nd conductivity	oounds, single in extended s	molecule magi ystems. Applic	nets. Electrical	
Compulsory entrance requirements							
Recommended participation require and/or individual courses of the mo		or the module					
Language(s) of instruction and exam	ination		German or English				
Weight of the module grade in the overall grade			Not graded				
Frequency of module offer			Only in the win	ter term			
Reasons for compulsory attendance							
Person responsible for the module			UnivProf. Dr. Eva Rentschler				
			0				
Transferability of the module to oth	er degree	programs	Master of Scier	nce Chemistry			



Creditpoints (LP) and workload 6 LP = 180 h Module duration (according to course plan) 1 Semester Courses/ Learning formats Type Regular term (Summer term) Mandatory/ elective Contact Time (SWH) Self Study Creditpoint a) Lecture "Supramolecular Catalysis" L 2 (1) M 3 103,5 h 4,5 b) Supporting exercise to a) E 2 (1) M 1 34,5 h 1,5 in order to complete the module, you have to fulfil the following requirements: Compulsory Attendance According to § 5 para. 3 Coursework Module examination Usually written exam (120 min), alternatively oral exam (30 min) Qualification Goals, learning outcome, competences The students - can reproduce knowledge on the terms mentioned in a structured way, . can establish connections and links between topics and contents within the subject and with related sub- disciplines, . can with related sub- disciplines, . Use of supramolecular interactions as control parameters to control catalytic processes. For homogeneous systems, thermal and light-driven catalyses are discussed with emphasison the correlation between catalysis structure, supramolecular interactions and resulting catalytic activity. Catalysis mechanisms including enantioselective catalysis, catalysis, under spatial confinement, inhibition requirements) Recommended partici	Module SK	Supram	olecular Cataly	[Modul-ł	[Modul-Kennnummer]			
Module duration (according to course plan) 1 Semester Courses/ Learning formats Type Regular term Winter term (Summer term) Mandatory/ elective (Summer term) Contact Time (SWH) Self Study Creditpoint a) Lecture ,Supramolecular Catalysis" L 2 (1) M 3 103,5 h 4,5 b) Supporting exercise to a) E 2 (1) M 1 34,5 h 1s ⁵ n order to complete the module, you have to fulfil the following requirements: Compulsory Attendance According to § 5 para. 3 Coursework Coursework Module examination Module examination Usually written exam (120 min), alternatively oral exam (30 min) Qualification Goals, learning outcome, competences Module examination Usually written exam (120 min), alternatively oral exam (30 min) Qualification Goals, learning outcome, competences E Coursework Coursework within the subject and with related sub- disciplines, c can describe the basic concepts and methods, c can work out and deepen partial contents independently. c can work out and deepen partial contents independently. c can transfer the learned contents to unknown tasks Coursework Set supramolecular interactions as control parameters to control catalytic processes. For homogeneous systems, thermal and light-driven catalyses are discussed with emphasistorin the correlation between catalysis structure, supramolecul	Mandatory or elective Module	Elective						
I semester Courses/ Learning formats Type Regular term (summer term) Mandator// elective Contact Time (SWH) Self Study Creditpoint a) Lecture "Supramolecular Catalysis" L 2 (1) M 3 103,5 h 4,5 b) Supporting exercise to a) E 2 (1) M 1 34,5 h 1,5 In order to complete the module, you have to fulfil the following requirements: Coursework Coursework Coursework Module examination According to § 5 para. 3 Coursework Coursework Coursework Module examination Usually written exam (120 min), alternatively oral exam (30 min) Qualification Goals, learning outcome, competences The students c can describe the basic concepts and methods, c can describe the basic concepts and methods, c can establish connections and links between topics and contents within the subject and with related sub- disciplines, c can transfer the learned contents to unknown tasks Courses Contents Use of supramolecular interactions as control parameters to control catalytic processes. For homogeneous systems, thermal and light-driven catalyses are discussed with emphasis on the correlation between catalyst study alysis under spatial confinement, inhibition and feedback loops, and autocatalysis are discussed. Contents Use of supramolecular interactions as control parameters to control ca	Creditpoints (LP) and workload	6 LP = 18	0 h					
Courses/ Learning formatsTypeWiner starting in Winner term (Summer term)Mandatory/ electiveContactSelf StudyCreditpointa) Lecture "Supramolecular Catalysis"L2 (1)M3103,5 h4,5b) Supporting exercise to a)E2 (1)M134,5 h1,5In order to complete the module, you have to fulfil the following requirements:Image: Compulsory AttendanceImage: Compulsory AttendanceActive participationAccording to § 5 para. 3Image: Compulsory AttendanceImage: Computer co	Module duration (according to course plan)	1 Semest	er					
Catalysis" L 2 (1) MI 3 103,5 fl 4,5 b) Supporting exercise to a) E 2 (1) M 1 34,5 fl 1,5 b) Supporting exercise to a) E 2 (1) M 1 34,5 fl 1,5 In order to complete the module, you have to fulfil the following requirements: Compulsory Attendance Active participation According to § 5 para. 3 Coursework According to § 5 para. 3 Coursework Module examination Usually written exam (120 min), alternatively oral exam (30 min) Qualification Goals, learning outcome, competences The students Can describe the basic concepts and methods, Can establish connections and links between topics and contents within the subject and with related sub-disciplines, Can establish connections and links between topics and contents within the subject and with related sub-disciplines, can transfer the learned contents to unknown tasks Contents Use of supramolecular interactions as control parameters to control catalytic processes. For homogeneous systems, thermal and light-driven catalyses are discussed with emphasis on the correlation between catalysis, catalysis, catalysis under spatial confinement, inhibition and feedback loops, and autocatalysis are discussed. For heterogeneous systems, supramolecular interactions and solids, e.g. metal organic frameworks (MOFs) are discussed. Compulsory entrance re		Туре	when starting in Winter term			Self Study	Creditpoints	
In order to complete the module, you have to fulfil the following requirements: Compulsory Attendance Active participation According to § 5 para. 3 Coursework Module examination Usually written exam (120 min), alternatively oral exam (30 min) Qualification Goals, learning outcome, competences The students Can reproduce knowledge on the terms mentioned in a structured way, Can describe the basic concepts and methods, Can work out and deepen partial contents independently, Can establish connections and links between topics and contents within the subject and with related sub- disciplines, Can transfer the learned contents to unknown tasks Contents Use of supramolecular interactions as control parameters to control catalytic processes. For homogeneous systems, thermal and light-driven catalyses are discussed with emphasis on the correlation between catalysts structure, supramolecular interactions and resulting catalytic, activity. Catalysis mechanisms including enantioselective catalysis, under spatial confinement, inhibition and feedback loops, and autocatalysis are discussed. For heterogeneous systems, supramolecular effects in colloids, polymers and solids, e.g. metal organic frameworks (MOFs) are discussed. Compulsory entrance requirements Recommended participation requirement(s) for the module and/or individual courses of the module Language(s) of instruction and examination German or English Weight of the module ot other degree programs Master of Science Chemistry	a) Lecture "Supramolecular Catalysis"	L	2 (1)	М	3	103,5 h	4,5	
Compulsory Attendance Active participation According to § 5 para. 3 Coursework Module examination Usually written exam (120 min), alternatively oral exam (30 min) Qualification Goals, learning outcome, competences The students • can reproduce knowledge on the terms mentioned in a structured way, • can describe the basic concepts and methods, • can work out and deepen partial contents independently, • can establish connections and links between topics and contents within the subject and with related sub- disciplines, • can transfer the learned contents to unknown tasks Contents Use of supramolecular interactions as control parameters to control catalytic processes. For homogeneous systems, thermal and light-driven catalyses are discussed with emphasis on the correlation between catalyst structure, supramolecular interactions and resulting catalytic activity. Catalysis mechanisms including enantioselective catalysis, catalysis under spatial confinement, inhibition and feedback loops, and autocatalysis are discussed. For heterogeneous systems, supramolecular effects in colloids, polymers and solids, e.g. metal organic frameworks (MOFs) are discussed. Compulsory entrance requirements German or English Weight of the module grade in the overall grade Not graded Frequency of module offer Only in the summer term Reasons for compulsory attendance Person responsible for the module Person responsible for the module Univ-Prof. Dr. Carsten Streb	b) Supporting exercise to a)	E	2 (1)	М	1	34,5 h	1,5	
Active participation According to § 5 para. 3 Coursework Usually written exam (120 min), alternatively oral exam (30 min) Qualification Goals, learning outcome, competences The students • can reproduce knowledge on the terms mentioned in a structured way, • can describe the basic concepts and methods, • can work out and deepen partial contents independently, • can establish connections and links between topics and contents within the subject and with related sub- disciplines, • can transfer the learned contents to unknown tasks Contents Use of supramolecular interactions as control parameters to control catalytic processes. For homogeneous systems, thermal and light-driven catalyses are discussed with emphasis on the correlation between catalysis, catalysis under spatial confinement, inhibition and feedback loops, and autocatalysis are discussed. For heterogeneous systems, supramolecular effects in colloids, polymers and solids, e.g. metal organic frameworks (MOFs) are discussed. Compulsory entrance requirements Recommended participation requirement(s) for the module and/or individual courses of the module And/or individual courses of the module Not graded Frequency of module effer Only in the summer term Reasons for compulsory attendance Person responsible for the module to other degree programs	In order to complete the module, y	ou have to	fulfil the following	requirements:			•	
Coursework Usually written exam (120 min), alternatively oral exam (30 min) Qualification Goals, learning outcome, competences The students • can reproduce knowledge on the terms mentioned in a structured way, • can describe the basic concepts and methods, • can work out and deepen partial contents independently, • can stablish connections and links between topics and contents within the subject and with related sub- disciplines, • can transfer the learned contents to unknown tasks • can transfer the learned contents to unknown tasks Contents • Use of supramolecular interactions as control parameters to control catalytic processes. For homogeneous systems, thermal and light-driven catalyses are discussed with emphasis on the correlation between catalysis, catalysis under spatial confinement, inhibition and feedback loops, and autocatalysis are discussed. For heterogeneous systems, supramolecular effects in colloids, polymers and solids, e.g. metal organic frameworks (MOFs) are discussed. Compulsory entrance requirements Recommended participation requirement(s) for the module and/or individual courses of the module Language(s) of instruction and examination German or English Weight of the module offer Only in the summer term Reasons for compulsory attendance Person responsible for the module Person responsible for the module UnivProf. Dr. Carsten Streb Transferability of the module to other degree programs Master of Science Chemistry <td>Compulsory Attendance</td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td>	Compulsory Attendance							
Module examination Usually written exam (120 min), alternatively oral exam (30 min) Qualification Goals, learning outcome, competences The students can reproduce knowledge on the terms mentioned in a structured way, can describe the basic concepts and methods, can work out and deepen partial contents independently, can establish connections and links between topics and contents within the subject and with related sub- disciplines, can transfer the learned contents to unknown tasks Contents Use of supramolecular interactions as control parameters to control catalytic processes. For homogeneous systems, thermal and light-driven catalyses are discussed with emphasis on the correlation between catalyst structure, supramolecular interactions and resulting catalytic activity. Catalysis mechanisms including enantioselective catalysis, catalysis under spatial confinement, inhibition and feedback loops, and autocatalysis are discussed. For homogeneous systems, supramolecular effects in colloids, polymers and solids, e.g. metal organic frameworks (MOFs) are discussed. Compulsory entrance requirements Image of instruction and examination German or English Mot graded Frequency of module offer Only in the summer term Reasons for compulsory attendance Image of instruction and examination Person responsible for the module UnivProf. Dr. Carsten Streb Transferability of the modul	Active participation	According	g to § 5 para. 3					
Qualification Goals, learning outcome, competences The students • can reproduce knowledge on the terms mentioned in a structured way, • can describe the basic concepts and methods, • can work out and deepen partial contents independently, • can establish connections and links between topics and contents within the subject and with related sub- disciplines, • can transfer the learned contents to unknown tasks. Contents Use of supramolecular interactions as control parameters to control catalytic processes. For homogeneous systems, thermal and light-driven catalyses are discussed with emphasis on the correlation between catalyst structure, supramolecular interactions and resulting catalytic activity. Catalysis mechanisms including enantioselective catalysis, catalysis under spatial confinement, inhibition and feedback loops, and autocatalysis are discussed. For heterogeneous systems, supramolecular effects in colloids, polymers and solids, e.g. metal organic frameworks (MOFs) are discussed. Compulsory entrance requirement(s) for the module and/or individual courses of the module Language(s) of instruction and examination German or English Weight of the module grade in the overall grade Not graded Frequency of module offer Only in the summer term Reasons for compulsory attendance Person responsible for the module Person responsible for the module UnivProf. Dr. Carsten Streb Transferability of the module to o	Coursework					0		
The students • can reproduce knowledge on the terms mentioned in a structured way, • can describe the basic concepts and methods, • can work out and deepen partial contents independently, • can establish connections and links between topics and contents within the subject and with related sub- disciplines, • can establish connections and links between topics and contents within the subject and with related sub- disciplines, Contents • can transfer the learned contents to unknown tasks. Contents • Contents as control parameters to control catalytic processes. For homogeneous systems, thermal and light-driven catalyses are discussed with emphasis on the correlation between catalyst structure, supramolecular interactions and resulting catalytic activity. Catalysis mechanisms including enantioselective catalysis, catalysis under spatial confinement, inhibition and feedback loops, and autocatalysis are discussed. For heterogeneous systems, supramolecular effects in colloids, polymers and solids, e.g. metal organic frameworks (MOFs) are discussed. Compulsory entrance requirements • Recommended participation requirement(s) for the module and/or individual courses of the module Language(s) of instruction and examination German or English Weight of the module grade in the overall grade Not graded Frequency of module offer Only in the summer term Reasons for compulsory attendance • Only in the summer term Person responsible for the module UnivProf. Dr. Carsten Streb	Module examination	Usually w	ritten exam (120 m	in), alternatively	y oral exam (3	0 min)		
 can reproduce knowledge on the terms mentioned in a structured way, can describe the basic concepts and methods, can work out and deepen partial contents independently, can establish connections and links between topics and contents within the subject and with related sub- disciplines, can transfer the learned contents to unknown tasks. Contents Use of supramolecular interactions as control parameters to control catalytic processes. For homogeneous systems, thermal and light-driven catalyses are discussed with emphasis on the correlation between catalyst structure, supramolecular interactions and resulting catalytic activity. Catalysis mechanisms including enantioselective catalysis, catalysis under spatial confinement, inhibition and feedback loops, and autocatalysis are discussed. For heterogeneous systems, supramolecular effects in colloids, polymers and solids, e.g. metal organic frameworks (MOFs) are discussed. Compulsory entrance requirements Recommended participation requirement(s) for the module and/or individual courses of the module Language(s) of instruction and examination German or English Weight of the module grade in the overall grade Not graded Frequency of module offer Only in the summer term Reasons for compulsory attendance Person responsible for the module UnivProf. Dr. Carsten Streb Transferability of the module to other degree programs Master of Science Chemistry	Qualification Goals, learning outcom	ne, compe	tences					
and light-driven catalyses are discussed with emphasis on the correlation between catalyst structure, supramolecular interactions and resulting catalytic activity. Catalysis mechanisms including enantioselective catalysis, catalysis under spatial confinement, inhibition and feedback loops, and autocatalysis are discussed. For heterogeneous systems, supramolecular effects in colloids, polymers and solids, e.g. metal organic frameworks (MOFs) are discussed. Compulsory entrance requirements Recommended participation requirement(s) for the module and/or individual courses of the module Language(s) of instruction and examination Weight of the module grade in the overall grade Frequency of module offer Reasons for compulsory attendance Person responsible for the module Transferability of the module to other degree programs Master of Science Chemistry	 can establish connect disciplines, 	ions and lir	iks between topics a	and contents wi	thin the subje	ct and with rela	ated sub-	
Recommended participation requirement(s) for the module and/or individual courses of the moduleGerman or EnglishLanguage(s) of instruction and examinationGerman or EnglishWeight of the module grade in the overall gradeNot gradedFrequency of module offerOnly in the summer termReasons for compulsory attendanceUnivProf. Dr. Carsten StrebTransferability of the module to other degree programsMaster of Science Chemistry	and light-driven catalyses are discus interactions and resulting catalytic a confinement, inhibition and feedbac	sed with er ctivity. Cat ck loops, an	nphasis on the corre alysis mechanisms i d autocatalysis are	elation betweer ncluding enantion discussed. For h	n catalyst struc oselective cata neterogeneous	ture, supramo alysis, catalysis	lecular under spatial	
and/or individual courses of the moduleGerman or EnglishLanguage(s) of instruction and examinationGerman or EnglishWeight of the module grade in the overall gradeNot gradedFrequency of module offerOnly in the summer termReasons for compulsory attendanceUnivProf. Dr. Carsten StrebPerson responsible for the module to other degree programsMaster of Science Chemistry	Compulsory entrance requirements	5						
Language(s) of instruction and examination German or English Weight of the module grade in the overall grade Not graded Frequency of module offer Only in the summer term Reasons for compulsory attendance UnivProf. Dr. Carsten Streb Person responsible for the module to other degree programs Master of Science Chemistry			or the module					
Frequency of module offer Only in the summer term Reasons for compulsory attendance UnivProf. Dr. Carsten Streb Person responsible for the module UnivProf. Dr. Carsten Streb Transferability of the module to other degree programs Master of Science Chemistry				German or Eng	lish			
Reasons for compulsory attendance UnivProf. Dr. Carsten Streb Person responsible for the module UnivProf. Dr. Carsten Streb Transferability of the module to other degree programs Master of Science Chemistry	Weight of the module grade in the overall grade			Not graded				
Person responsible for the module UnivProf. Dr. Carsten Streb Transferability of the module to other degree programs Master of Science Chemistry	Frequency of module offer			Only in the sun	nmer term			
Transferability of the module to other degree programs Master of Science Chemistry	Reasons for compulsory attendance	2						
	Person responsible for the module			UnivProf. Dr.	Carsten Streb			
Other	Transferability of the module to oth	ner degree	programs	Master of Scier	nce Chemistry			
	Other							

Module MPC	Molecu	lar Photochemi	stry		[Modul-K	(ennnummer]	
Mandatory or elective Module	Elective				<u> </u>		
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	1 Semester					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Lecture "Molecular Photochemistry"	L	1 (2)	М	3	103,5 h	4,5	
b) Supporting exercise to a)	E	1 (2)	М	1	34,5 h	1,5	
In order to complete the module, yo	u have to	fulfil the following	requirements:				
Compulsory Attendance							
Active participation	According	to § 5 para. 3					
Coursework							
Module examination	Usually w	Usually written exam (120 min), alternatively oral exam (30 min)					
Qualification Goals, learning outcom	e, compet	ences					
 can reproduce knowled can describe the basic can work out and deep can establish connection disciplines, are able to transfer the gain a comprehensive of fundamental concepts 	concepts a ben partial ons and lin e contents overview o	and methods, contents independ ks between topics a they have learned of the cross-section	ently, and contents wi to unfamiliar ta: al discipline of p	thin the subjectsks,			
Contents							
chromophores, photokinetics, optical photosynthesis, photochemical probe	Electron transfer, fundamentals of photochemistry, photophysics and photochemistry of metal complexes and organic chromophores, photokinetics, optical spectroscopy, photocatalysis, solar energy conversion, natural and artificial photosynthesis, photochemical probes, supramolecular photochemistry, organic photoreactions, isomerisations, rearrangements, fragmentations, photochemistry in biological systems.						
Compulsory entrance requirements							
Recommended participation require and/or individual courses of the modest the modest set the mo		or the module					
Language(s) of instruction and exami	ination		German or Eng	lish			
Weight of the module grade in the o	verall grad	le	Not graded				
Frequency of module offer			Only in the win	ter term			
Reasons for compulsory attendance							
Person responsible for the module			UnivProf. Dr.	Katja Heinze			

Master of Science Chemistry

Other

Transferability of the module to other degree programs

Module Handbook	Version	As of	JG
Master of Science Biomedical Chemistry	4.0	03.04.2025	JGIO

Module FMM		ed Laboratory C	Course on Fu	nctional	[Modul-I	Kennnummer]	
	Molecu	lar Materials					
Mandatory or elective Module	Elective						
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	er					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Lab Course "Functional Molecular Materials"	APr	2 (1)	М	9	40,5 h	4,5	
b) Supporting Seminar to a)	S	2 (1)	М	1	34,5 h	1,5	
In order to complete the module, yo	u have to	fulfil the following	requirements:				
Compulsory Attendance	APr, S						
Active participation	According to § 5 para. 3						
Coursework							
Module examination				\O			
Qualification Goals, learning outcom	e, compe	tences					
evaluate them and ass are proficient in the th according to the rules are able to handle haze environmental regulat Contents	eoretical I of good sc ardous sul	background of their ientific practice	\sim				
		of functional materia			h		
Conducting experiments to elaborate and analytical methods, e.g. investiga experiments, determination of turno experimental results with theoretical	ntion of ele ver curves	ectronic and magne of catalyses or pho	tic properties, lu	uminescence c	or time-resolve	d spectroscopic	
Compulsory entrance requirements							
Recommended participation require and/or individual courses of the modest and the modest courses of the modest and the modest		or the module	Modules "Molecular Photochemistry", "Supramolecular Catalysis" und "Electrons in Molecules"				
Language(s) of instruction and exam	ination		German or English				
Weight of the module grade in the o	verall grad	de	Not graded				
Frequency of module offer			Only in the summer term				
Reasons for compulsory attendance			According to HochSchG § 26 Abs. 2 (7), internship; internship-accompanying upper seminar according to § 5 Abs. 5: discussion of safety-relevant details of and discussion of internship experiments.				
Person responsible for the module			UnivProf. Dr.	Carsten Streb			
Transferability of the module to othe	er degree	programs	Master of Scier	nce Chemistry			
Other							



Module SpA	Trace Analysis I [Modul-Kennnum					(ennnummer]
Mandatory or elective Module	Elective					
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration (according to course plan)	1 Semest					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints
a) Lecture "Inorganic Trace and Species Analysis"	L	1 (2)	М	2	69 h	3
b) Lecture "Organic Trace Analysis"	L	1 (2)	М	2	69 h	3
In order to complete the module, yo	ou have to	fulfil the following	requirements:			
Compulsory Attendance						
Active participation	Gemäß § 5 Abs. 3					
Coursework					0	
Module examination	Usually w and b)	ritten exam (120 m	in), alternatively	y oral exam (3	0 min) on the c	ontents of a)
Qualification Goals, learning outcon	ne, compe	tences				
lectures on organic trace analysis and knowledge. The students are able to: • reproduce principles f • identify the main area analysis, species analy • relate keywords such detection to the meth • evaluate analytical met to a set trace analytica • to understand the exp critically evaluate this	for the separation of the separation of applications, medicates as food safe to determine the second	aration and detectic ation of analysis, su al and diagnostic an ety or water contan select and develop	on of organic an ch as environme alysis nination, doping suitable instrum	d inorganic an ental analysis, g tests, genetic nental methoo	alytes technical and in c analysis or aut ds and procedu	ndustrial thenticity res according
Contents a) Sampling of organic analytes, prece electrophoretic separation technique of organic mass spectrometry, ionisa (bioanalysis, environmental analysis, b) Physical fundamentals of atomic s resolution AAS, atomic emission spec discharges, microwave plasmas, lase ray fluorescence analysis.	es, bioanal ation techn forensic a pectromet ctrometry	ytical separation teo iques, mass spectro nalysis). ry, atomic absorptic with flames and plas	chniques, miniat metric analyser on spectrometry smas, sample in	turisation of se s, applied orga r, mono-/polya troduction teo	eparation techr anic trace analy chromators, de chniques, arc ar	iques, basics sis tectors, high- nd spark
Compulsory entrance requirements						
Recommended participation require		or the module				
Recommended participation require and/or individual courses of the mo	dule	or the module	German or Eng	lish		
Recommended participation require and/or individual courses of the mo Language(s) of instruction and exam	odule		German or Eng Not graded	lish		
Recommended participation require and/or individual courses of the mo Language(s) of instruction and exam Weight of the module grade in the c	odule		-			
Recommended participation require and/or individual courses of the mo Language(s) of instruction and exam Weight of the module grade in the o Frequency of module offer	dule nination overall grad		Not graded			
Compulsory entrance requirements Recommended participation require and/or individual courses of the mo Language(s) of instruction and exam Weight of the module grade in the of Frequency of module offer Reasons for compulsory attendance Person responsible for the module	dule nination overall grad		Not graded Only in the win	ter term		
Recommended participation require and/or individual courses of the mo Language(s) of instruction and exam Weight of the module grade in the o Frequency of module offer	dule nination overall grad	de	Not graded	ter term olas. H. Bings		



Module SpaP	Trace A	Trace Analysis II [Modul-Kennnummer]					
Mandatory or elective Module	Elective	Elective					
Creditpoints (LP) and workload	6 LP = 18	5 LP = 180 h					
Module duration (according to course plan)	1 Semest	er					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Practical course "Trace Analysis II"	APr	2 (1)	М	4	78 h	4	
b) Supporting Seminar to a)	Seminar	2 (1)	М	2	39 h	2	
In order to complete the module, yo	order to complete the module, you have to fulfill the following requirements:						
Compulsory Attendance	APr, S					V	
Active participation	According	to § 5 para. 3			~		
Coursework							
Module examination				-	0		
Qualification Goals, learning outcom	e, compet	ences					

Building on analytical contents and working techniques already learned in the Bachelor's degree programme, students acquire special expertise in the field of advanced instrumental trace analysis in the module Instrumental Trace Analysis II (practical course). The contents are developed, deepened and practically implemented in the form of an advanced practical course on organic trace analysis and elemental analysis and a lecture seminar. Newly acquired knowledge is always integrated into the existing knowledge. The students acquire in-depth knowledge of the current methods of instrumental trace analysis (chromatography, atomic spectrometry, molecular spectroscopy, mass spectrometry).

The students are able to:

- apply advanced analytical-instrumental working techniques
- statistically evaluate recorded measurement data
- carry out trace analysis work independently and on their own responsibility
- scientifically record, interpret and present the results of their experiments
- agree on individual work steps when working in groups of two, to plan them together and to implement them in a coordinated manner
- realise demanding research-related experiments in parallel within a time window (self-, time- and resource management)
- analyse and evaluate current scientific literature
 - independently prepare and present a scientific Presentation on a (given) current analytical-chemical topic.

Contents

a) Experiments in groups of two on the determination of organic analytes by means of GC-MS and HPLC-MS (mode of operation, set-up, column types, ionisation techniques, detectors, analysers, MS/MS, derivatisation), by means of ambient MS (set-up and mode of operation of corresponding ion sources, advantages and disadvantages, areas of application), and by means of aerosol mass spectrometry (AMS). Experiments in groups of two on inorganic trace analysis based on analyte samples of different matrices by means of mass and emission spectrometry in connection with inductively coupled plasma (ICP-OES, ICP-MS) and X-ray spectroscopy (TXRF). Consideration of different sample preparation/digestion methods and systems of sample introduction.

b) Current analytical-chemical topics are discussed. The students independently prepare a scientific presentation on one of these given topics and present it within the framework of the seminar. Independent research and evaluation of relevant literature are important.

Compulsory entrance requirements	Module "Trace Analysis I"
Recommended participation requirement(s) for the module and/or individual courses of the module	
Language(s) of instruction and examination	German or English
Weight of the module grade in the overall grade	Not graded
Frequency of module offer	Only in the summer term

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Module MC1	Macromolecular Chemistry [Modul-Kennnummer]					ennnummer]
Mandatory or elective Module	Elective					
Creditpoints (LP) and workload	6 LP = 180 h					
Module duration (according to course plan)	1 Semest	er				
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints
a) Lecture Part 1: "Synthesis and use of polymers". Part 2: "Physical Chemistry of Polymers".	L	1 - 3 (1 - 3)	М	3	103,5 h	4,5
b) Supporting exercise to a)	E	1 - 3 (1 - 3)	М	1	34,5 h	1,5
In order to complete the module, yo	ou have to	fulfil the following	requirements:			
Compulsory Attendance				s (0	
Active participation	b) Accord	ing to § 5 para. 3 (u	sually exercise a	assignments)		
Coursework						
Module examination	Usually w	ritten exam (120 m	in), alternativel	y oral exam (30) min)	
Qualification Goals, learning outcon	ne, compe	tences				
comparison to other n acquire the basics of p critically evaluate poly regard to the respectiv get to know basic char questions conceptually understa	oolymer ch vmerisatior ve limitatic racterisatic und and qu	emistry, types of po methods, both wit ons concerning poly on methods and to e antitatively discuss	lymerisation, ch h regard to the dispersity, evaluate them w the structure ar	nain and step g achievable mo vith regard to t nd dynamics of	blecular weights heir suitability	for specific
thermodynamically de	escribe mad	cromolecular multi-	substance syste	ms.		
Part 1: General basics: tasks of polymer scie Polymer synthesis: Polycondensation Radical and ionic methods of polyme polyinsertion Polymerisation, polyinsertion, cataly Polymerisation in heterophase (emu Polymer modification: cellulose, rubl Controlled and living polymerisation Part 2: Polymer structure: block copolymers statistics, entropy elasticity, Flory exp	n (step grov er synthesis sts (initiato lsion, dispe per, polym processes, c, conforma ponent and ers in solut	wth), Carothers equ s, kinetics, chain trai ers). ersion, suspension). er analogue reaction ring opening reaction stion of macromoled d scale laws.	ation, polymeris nsfer, copolyme ns. ons, solid phase cules, errant sta	erisation, cataly e synthesis. tistics, RIS mo	ytic polymerisat	
Molecular characterisation of polymo mass spectrometry, static light scatte Polymer dynamics: Rouse and Zimm Polymer thermodynamics: Flory-Hug Compulsory entrance requirements	model. gins theory					
mass spectrometry, static light scatte Polymer dynamics: Rouse and Zimm Polymer thermodynamics: Flory-Hug Compulsory entrance requirements Recommended participation require	model. gins theory ement(s) fo					
mass spectrometry, static light scatte Polymer dynamics: Rouse and Zimm Polymer thermodynamics: Flory-Hug Compulsory entrance requirements	model. gins theory ement(s) fo dule		English			

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Frequency of module offer	Every term
Reasons for compulsory attendance	
Person responsible for the module	Prof. Dr. Andreas Walther
Transferability of the module to other degree programs	Bachelor of Science Biomedical Chemistry, Bachelor of Science Chemistry, Master of Science Chemistry
Other	Recommended Literature: Tieke – Makromolekulare Chemie. Eine Einführung (Wiley Koltzenburg, Maskos, Nuyken – Polymere: Synthese, Eigenschaften und Anwendungen (Springer) Lechner, Gehrke, Nordmeier – Makromolekulare Chemie (Springer) Seiffert – Physical Chemistry of Polymers: A Conceptual Introduction (DeGruyter)
hormation	noute



Module MC1P	Practical Course Biomacromolecular Chemistry [Modul-Kennnummer							
Mandatory or elective Module	Elective	Elective						
Creditpoints (LP) and workload	6 LP = 18	6 LP = 180 h						
Module duration (according to course plan)	1 Semester							
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints		
a) Practical Course Biomacromolecular Chemistry for Advanced Students 1	APr	1 - 3 (1 - 3)	М	6	117 h	6		
In order to complete the module, yo	ou have to	fulfil the following	requirements:					
Compulsory Attendance	APr							
Active participation	Accordin	g to § 5 para. 3						
Coursework					2			
Module examination								
Qualification Goals, learning outcon	ne, compe	tences						
deal effectively with t within a defined time Contents Practical experiments are selected fr Experiments on polymer synthesis: r copolymerisation, polymerisation in biopolymers, silicones and biodegrad	window. om the fol adical poly heteropha	lowing areas: merisation, polycon se, networks; as we	idensation, livin	g/controlled p	olymerisation,	-		
Compulsory entrance requirements			Module "Macr	omolecular Ch	emistry			
Recommended participation require and/or individual courses of the mo	ement(s) fo	or the module			emistry			
Language(s) of instruction and exam	nination		German or English					
Weight of the module grade in the o	overall gra	de	Not graded					
Frequency of module offer			Every term					
Reasons for compulsory attendance			According to H	ochSchG § 26	Para. 2 (7), Pra	ctical Course		
Person responsible for the module			UnivProf. Dr.	Sebastian Seif	fert			
Transferability of the module to oth	er degree	programs	Bachelor of Sci	ence Biomedio	cal Chemistry			
Other								
10								

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Module MC2	Moder	n and Industrial	Aspects of P	olvmer	[Modul-k	(ennnummer]	
	Materia			,			
Mandatory or elective Module	Elective						
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	er					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Lecture Part 1: "Synthesis and Use of Polymer Materials" Part 2: "Physical Chemistry of Polymeric Materials	L	1 (2)	Μ	3	103,5 h	4,5	
 b) Seminar "Modern and Industrial Aspects of Polymer Materials" 	S	1 (2)	М	1	34,5 h	1,5	
In order to complete the module, yo	u have to	fulfil the following	requirements:				
Compulsory Attendance				.0			
Active participation	According	g to § 5 para. 3 (usua	ally successful p	resentation ir	the seminar)		
Coursework				2			
Module examination	Usually w	ritten exam (120 m	in), alternatively	y oral exam (3	0 min)		
Qualification Goals, learning outcom	e, compe	tences					
 describe central challe understand current re- elastomers, composite and bio-inspired mater describe the rheology phenomenologically, b reproduce the basic ch crystals. 	search que materials rial design of polyme oth qualit	estions of an acader , weak interactions , rs in the melt and so atively and quantita	nic nature: For e in polymer scier plution states m tively.	example, sequ nce, self-asser ethodological	ience control, tl nbly, responsiv ly, conceptually	hermoplastic e materials / and	
Contents							
Modern methods of polymer synthes Advanced composite r Responsive and switch Biomimetic concepts in Phase-segregated poly Polymer nanoparticles Fundamentals of rheology: Viscoelasticity Complex rheological m Time-temperature sup Rheology of polymer s transition.	naterials, I able mate n polymer "mer syste and self-a naterial pro erposition ystems: Re	rials science ms in application, th ssembled nanostru operties eptation in melt and	nermoplastic ela ctures solution, rubbe	er elasticity of	-	-	
Building on this: comprehensive and the state of melts, semi-dilute solution				ure, dynamics	and properties	s of polymers in	
Compulsory entrance requirements	montial	w the medule					
Recommended participation require and/or individual courses of the mod		or the module	Module "Macro	omolecular Ch	nemistry"		
Language(s) of instruction and exam	ination		English				
Weight of the module grade in the o	verall grad	de	Not graded				
Frequency of module offer			Only in the win	ter term			



Reasons for compulsory attendance	
Person responsible for the module	UnivProf. Dr. Andreas Walther
Transferability of the module to other degree programs	Master of Science Chemistry
Other	Recommended Literature: Koltzenburg, Maskos, Nuyken – Polymere: Synthese, Eigenschaften und Anwendungen (Springer) Lechner, Gehrke, Nordmeier – Makromolekulare Chemie (Springer) Rubinstein, Colby – Polymer Physics (Oxford University Press)
Nit	press)
Informatio	



Module MC3	Colloid	Chemistry and	Medical Poly	vmers	[Modul-H	(ennnummer]		
Mandatory or elective Module	Elective				<u>+</u>			
Creditpoints (LP) and workload	6 LP = 18	0 h						
Module duration (according to course plan)	1 Semes	1 Semester						
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints		
a) Lecture "Colloid Chemistry"	L	1 (2)	М	2	69 h	3		
b) Lecture "Medically relevant polymers"	L							
In order to complete the module, y	ou have to	fulfil the following	requirements:					
Compulsory Attendance								
Active participation	Accordin	g to § 5 para. 3						
Coursework					0			
Module examination	b) Usually	 a) Usually written exam (60 min), alternatively oral exam (30 min) b) Usually written exam (60 min), alternatively oral exam (30 min). Both exams must be passed, the module grade results from the arithmetic mean of bot exams. 						
Qualification Goals, learning outco	me, compe	tences						
An in-depth insight into the produc medical applications is provided. The students are able to: • reproduce and expla • discuss colloidal syst • work out and reprod	in methods ems with re	for the investigation gard to their charac	n of nanostructu teristic time, lei	ures and (poly ngth and energ	mer) surfaces, gy scales,			
Contents								
 a) Interfacial and colloid chemistry with different properties for differe b) Synthesis methods for materials biodegradation of polymeric materi (aliphatic polyesters, polyethylene) and vaccines; artificial extracellular 	nt application for use in m ials; biocom glycol, silico matrix mate	ons, characterisatio edicine, implants fo patibility and biode nes, polypeptides a	n. or dental applica gradability of po	ations or as pro olymer classes	ostheses; basic for medical ap	principles of plications		
Compulsory entrance requirement	s							
Recommended participation requirement(s) for the module and/or individual courses of the module			Module "Macromolecular Chemistry"					
		or the module	Module "Macr	omolecular Ch	emistry			
	odule	or the module	Module "Macr German or Eng		emistry			
and/or individual courses of the m	odule mination				emistry			
and/or individual courses of the m Language(s) of instruction and exa	odule mination		German or Eng	glish	emistry			
and/or individual courses of the m Language(s) of instruction and exa Weight of the module grade in the	odule mination overall gra		German or Eng Not graded	glish	emistry"			
and/or individual courses of the m Language(s) of instruction and exa Weight of the module grade in the Frequency of module offer	odule mination overall gra- e		German or Eng Not graded	ilish nter term	emistry			
and/or individual courses of the m Language(s) of instruction and exa Weight of the module grade in the Frequency of module offer Reasons for compulsory attendanc	odule mination overall gra	de	German or Eng Not graded Only in the wir	lish hter term Holger Frey	emistry			

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Module MC4	Complex (Supra)Molecular Systems and ^[M] Biopolymers						
Mandatory or elective Module	Elective				<u>+</u>		
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	er					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Lecture "Complex (Supra)Molecular Systems"	L	2 (1)	М	2	69 h	3	
b) Lecture "Biopolymers"	L	2 (1)	М	2	69 h	3	
In order to complete the module, y	ou have to	fulfil the following	requirements:				
Compulsory Attendance							
Active participation	According	g to § 5 para. 3			0		
Coursework							
Module examination	 a) Usually written exam (60 min), alternatively oral exam (30 min) b) Usually written exam (60 min), alternatively oral exam (30 min). Both exams must be passed, the module grade results from the arithmetic mean of bot exams. 						
Qualification Goals, learning outco	me, compet	tences					
 The students are able to: evaluate biologically assembly, understand and appl synthetic systems, understand and reprine Distinguish equilibriu Understand the basic 	y recognitio oduce biolo m and non-	n motifs, weak inte gical and chemical r equilibrium system	ractions and org reaction networ s.	ganisational pr ks and their d	inciples in natu ynamics.	iral and	
Contents		·					
 a) Supramolecular Chemistry and Su Networks and Systems; Non-equilib Adaptive and Interactive Materials. b) Polysaccharides (cellulose and de polyisoprenoids and natural rubber RNA); Proteins and scleroproteins (or 	rium States rivatives, ch); Nanocellu	; Chemical Reaction nitin, starch, glycoge lose/nanochitin/ba	Networks, Dyn en); Lignins; Pol cterial cellulose	amic DNA Nar yesters (polyh ; Polynucleoti	noscience, Dissi ydroxyalkanoat des in material	pative, tes), s context (DNA,	
Compulsory entrance requirement	s						
Recommended participation requinand/or individual courses of the m		or the module	Module "Macr	omolecular Ch	emistry"		
Language(s) of instruction and exa	mination		German or Eng	lish			
Weight of the module grade in the	overall grad	de	Not graded				
Frequency of module offer			Only in the sun	nmer term			
Reasons for compulsory attendanc	e						
Person responsible for the module			UnivProf. Dr.	Pol Besenius			
Transferability of the module to ot	her degree	programs	Master of Scier	nce Chemistry			



Module MMPC	Moder	n Methods of Pł	nysical Chem	Modern Methods of Physical Chemistry [Modul-Kennnumm					
Mandatory or elective Module	Elective	Elective							
Creditpoints (LP) and workload	6 LP = 18	0 h							
Module duration (according to course plan)	1 Semest	1 Semester							
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints			
a) Lecture "Modern Methods of Physical Chemistry"	L	1 o. 2 (1 o. 2)	Р	3	103,5h	4,5			
b) Supporting exercise to a)	E	1 o. 2 (1 o. 2)	Р	1	34,5 h	1,5			
In order to complete the module, y	ou have to	fulfil the following	requirements:						
Compulsory Attendance									
Active participation	Accordin	g to § 5 para. 3							
Coursework					0				
Module examination	Usually w	ritten exam (120 m	in), alternatively	/ oral exam (30) min)				
Qualification Goals, learning outco	me, compe	tences							
physical chemistry, u should be able to sele corresponding measu Contents a) Basics of modern microscopic me • Imaging microscopy meth	ect the appr arement dat thods with ods (confoc	opriate methods fo ta in order to succes examples from thei	or different expension ssfully get to the ir field of applica	rimental ques bottom of ne ation. Topics a	tions and inter w phenomena re for example	pret the			
 Current topics in modern Microscopy methods for the Modern methods for the 	he analysis characterisa	pectroscopy, e.g. si of dynamic process ation of molecular p	ngle molecule s es and intermol hysico-chemica	pectroscopy Fl ecular interact I parameters (RET cions (FRAP) NanoSPR)				
 Microscopy methods for t Modern methods for the b) In-depth or supplementary topics 	he analysis characterisa s from the le	pectroscopy, e.g. si of dynamic process ation of molecular p	ngle molecule s es and intermol hysico-chemica	pectroscopy Fl ecular interact I parameters (RET cions (FRAP) NanoSPR)				
 Microscopy methods for t Modern methods for the 	he analysis characterisa s from the le s rement(s) for	pectroscopy, e.g. si of dynamic process ation of molecular p ecture area with pra	ngle molecule s es and intermol hysico-chemica	pectroscopy Fl ecular interact I parameters (RET cions (FRAP) NanoSPR)				
 Microscopy methods for t Modern methods for the b) In-depth or supplementary topics Compulsory entrance requirement Recommended participation requirement 	he analysis characterisa s from the lo s rement(s) fo odule	pectroscopy, e.g. si of dynamic process ation of molecular p ecture area with pra	ngle molecule s es and intermol hysico-chemica	pectroscopy Fl ecular interact parameters (and applicatio	RET cions (FRAP) NanoSPR)				
 Microscopy methods for t Modern methods for the b) In-depth or supplementary topics Compulsory entrance requirement Recommended participation requirement 	he analysis characterisa s from the le s rement(s) for odule nination	pectroscopy, e.g. si of dynamic process ation of molecular p ecture area with pra or the module	ngle molecule s es and intermol hysico-chemica actical exercises German or Eng	pectroscopy Fl ecular interact parameters (and applicatio	RET cions (FRAP) NanoSPR)				
 Microscopy methods for t Modern methods for the b) In-depth or supplementary topics Compulsory entrance requirement Recommended participation requirement and/or individual courses of the methods Language(s) of instruction and example weight of the module grade in the 	he analysis characterisa s from the le s rement(s) for odule nination	pectroscopy, e.g. si of dynamic process ation of molecular p ecture area with pra or the module	ngle molecule s es and intermol hysico-chemica actical exercises	pectroscopy Fl ecular interact parameters (and applicatio	RET cions (FRAP) NanoSPR)				
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 Microscopy methods for t Modern methods for the b) In-depth or supplementary topics Compulsory entrance requirement Recommended participation require and/or individual courses of the methods Language(s) of instruction and example Weight of the module grade in the Frequency of module offer 	he analysis characterisa s from the los rement(s) fo odule mination overall gra	pectroscopy, e.g. si of dynamic process ation of molecular p ecture area with pra or the module	ngle molecule s es and intermol hysico-chemica actical exercises German or Eng Not graded	bectroscopy Fl ecular interact parameters (and application	RET cions (FRAP) NanoSPR)				
 Microscopy methods for t Modern methods for the b) In-depth or supplementary topics Compulsory entrance requirement Recommended participation requirement and/or individual courses of the methods Language(s) of instruction and example examples Weight of the module grade in the Frequency of module offer Reasons for compulsory attendance 	he analysis characterisa from the los rement(s) fo odule mination overall gra	pectroscopy, e.g. si of dynamic process ation of molecular p ecture area with pra- or the module de	ngle molecule s es and intermol hysico-chemica actical exercises German or Eng Not graded Every term	ectroscopy Fl ecular interact parameters (i and application lish	RET cions (FRAP) NanoSPR)				

Module MMPCP	Practic	al Course Mode	rn Methods o	of Spectros	Copy [Modul-K	ennnummer]
	and Mi	croscopy				
Mandatory or elective Module	Elective					
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration (according to course plan)	1 Semes	ter				
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints
a) Practical Course "Modern Methods of Spectroscopy and Microscopy"	Apr	1 o. 2 (1 o. 2)	М	3	103,5 h	4,5
b) Supporting seminar to a)	S	1 o. 2 (1 o. 2)	М	1	34,5 h	1,5
In order to complete the module, y	ou have to	fulfil the following	requirements:			
Compulsory Attendance	APr				N	
Active participation	According	g to § 5 para. 3				
Coursework						
Module examination						
Qualification Goals, learning outco	me, compe	tences				
on it according to sci critically assessed. Contents			iounus, their ov			
6-8 practical experiments from the	field of exp					
 time-resolved fluores confocal fluorescenc scanning tunneling n light microscopy transmission electroi Synthesis of CdSe na FRAP (fluorescence r Topics for the oral presentation are 	scence and e microscop nicroscopy n microscop nocrystals ecovery afte	electronic energy tra by and single molecu Y er photobleaching)	ansfer Ile microscopy			
 confocal fluorescenc scanning tunneling n light microscopy transmission electroi Synthesis of CdSe na FRAP (fluorescence r 	scence and e microscopy nicroscopy n microscop nocrystals ecovery afte chosen fro	electronic energy tra by and single molecu Y er photobleaching)	ansfer Ile microscopy			
 confocal fluorescenc scanning tunneling n light microscopy transmission electron Synthesis of CdSe na FRAP (fluorescence n Topics for the oral presentation are 	scence and e microscopy n microscopy nocrystals ecovery after chosen fro s rement(s) for	electronic energy tra y and single molecu y er photobleaching) m the field of practio	ansfer Ile microscopy			
 confocal fluorescence scanning tunneling m light microscopy transmission electron Synthesis of CdSe na FRAP (fluorescence r Topics for the oral presentation are Compulsory entrance requirement Recommended participation requi 	scence and e microscopy nicroscopy nocrystals ecovery after chosen fro s rement(s) for odule	electronic energy tra y and single molecu y er photobleaching) m the field of practio	ansfer Ile microscopy	and related a		
 confocal fluorescence scanning tunneling n light microscopy transmission electron Synthesis of CdSe na FRAP (fluorescence r Topics for the oral presentation are Compulsory entrance requirement Recommended participation requia and/or individual courses of the m 	scence and e microscopy n microscopy nocrystals ecovery aft chosen fro s rement(s) fo odule mination	electronic energy tra by and single molecu y er photobleaching) m the field of praction or the module	ansfer Ile microscopy cal experiments	and related a		
 confocal fluorescence scanning tunneling n light microscopy transmission electron Synthesis of CdSe na FRAP (fluorescence n Topics for the oral presentation are Compulsory entrance requirement Recommended participation requiand/or individual courses of the m Language(s) of instruction and examination 	scence and e microscopy n microscopy nocrystals ecovery aft chosen fro s rement(s) fo odule mination	electronic energy tra by and single molecu y er photobleaching) m the field of praction or the module	ansfer Ile microscopy cal experiments German or Eng	and related a		
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	Introdu	ction in Nuclea	<mark>r Chemistry</mark>		[Modul-K	(ennnummer]	
Mandatory or elective Module	Elective						
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	er					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Lecture "Introduction in Nuclear Chemistry"	L	1 o. 2 (1 o. 2)	М	2	69 h	3	
b) Supporting exercise to a)	Е	1 o. 2 (1 o. 2)	М	1	34,5 h	1,5	
c) Supporting Seminar to a)	S	1 o. 2 (1 o. 2)	М	1	34,5	1,5	
In order to complete the module, yo	u have to	fulfil the following	requirements:			•	
Compulsory Attendance	S						
Active participation	According	g to § 5 para. 3					
Coursework							
Module examination	Usually w	ritten exam (120 m	in), alternatively	/ oral exam (3	0 min)		
Qualification Goals, learning outcom	ne, compe	tences					
Contents							
a) History of radioactivity / structure liquid droplet model and shell model transformations, units of radioactivit radioactivity, spontaneous fission / s Annihilation, X-rays, Auger electrons measurement of nuclear radiation: d compound nuclei, heavy ion reaction b) In the exercises, exercise assignme c) Presentations will be given on topi determination; Discovery and proper environment; The tracer principle an application of radionuclides in life sci	/ instabilit y, natural econdary c / interacti ifferent typ s, high ene ents are ca cs that cor ties of the d its applie	cy of nuclei and nuc adionuclides / prim conversions: electro on with matter: pho bes of detectors / n ergy reactions, indu- lculated. nplement the lectur neutron; Discovery cations in chemistry	lear transforma hary transforma magnetic transi otoelectric effec uclear reactions ced fission. re content, e.g of nuclear fissio and medicine;	tion principles tions: α -conver tions, convers t, Compton ef : Energetics, c : α -/ β -/ γ -spec on; Natural rac Particle accele	/ mathematica rsion, β-conver ion electrons / fect, pair forma ross section, di trometry; Radic dioactivity in th rators; Product	al relations of rsion, cluster post effects: ation / rect reactions ometric age e ion and	
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Transferability of the module to other degree programs	Bachelor of Science Biomedicinal Chemistry, Bachelor of Science Chemistry, Bachelor of Science Geoscience, Master of Science Chemistry, Master of Science Physics
Other	 Recommended Literature: JV. Kratz, K. H. Lieser: Nuclear and Radiochemistry, Wiley-VCH, 2013 F. Rösch: Nuclear and Radiochemistry, De Gruyter, 2014 Vértes, S. Nagy, Z. Klencsár, R. G. Lovas, F. Rösch (Eds.), Handbook of Nuclear Chemistry, Springer 2011
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Module KCP	Lab Course Nuclear Chemistry 1 [Modul-Kennnummer]		(ennnummer]				
Mandatory or elective Module	Elective				<u>_</u>		
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semester						
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Lab Course "Nuclear Chemistry 1"	APr	1 o. 2 (1 o. 2)	М	6	72 h	4,5	
b) Supporting Seminar to a)	S	1 o. 2 (1 o. 2)	М	1	34 <i>,</i> 5h	1,5	
In order to complete the module, yo	u have to	fulfil the following	requirements:				
Compulsory Attendance	APr, S				X	0	
Active participation	According	to § 5 para. 3					
Coursework							
Module examination	Oral exam	n (30 minutes, not g	raded)	5	0		
Qualification Goals, learning outcom	e, compet	ences					
Contents Production and handling of radioactiv equilibrium, interaction of radiation v emission tomography, nuclear reaction behaviour of neptunium.	vith matte	r, gamma spectroso	copy, dosimetry	and radiation	protection, ba	sics of positron	
Compulsory entrance requirements			Module "Intro	duction in nucl	lear Chemistry'	u and a start and a start a sta	
Recommended participation require and/or individual courses of the modest terms and the modest of th		or the module					
Language(s) of instruction and exam	ination		German				
Weight of the module grade in the o	verall grad	le	Not graded				
Frequency of module offer			Every term				
Reasons for compulsory attendance			In accordance with HochSchG § 26 para. 2 (7), practical course; seminar accompanying practical course in accordance with § 5 para. 5: discussion of safety-relevant details of and discussion of practical course experiments.				
Person responsible for the module			UnivProf. Tho				
Transferability of the module to othe	er degree	programs	Bachelor of Science Biomedical Chemistry, Bachelor of Science Chemistry, Bachelor of Science Geoscience, Master of Science Chemistry, Master of Science Physics				
Other			 Recommended Literature: P. Hoffmann, K. H. Lieser: Methoden der Kern- und Radiochemie, VCH 1991 W. Stolz: Radioaktivität, Teubner, 2005 HG. Vogt, H. Schultz: Grundzüge des praktischen Strahlenschutzes, Hanser, 2011 				



Module QC1	Princip	les of Quantum	Chemistry		Integral	(ennnummer]
Mandatory or elective Module	Elective				<u>_</u>	
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration (according to course plan)	1 Semest	ter				
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoint
a) Lecture "Principles of Quantum Chemistry"	L	1 o. 3 (2)	М	3	103 <i>,</i> 5 h	4,5
b) Supporting exercise to a)	Е	1 o. 3 (2)	М	1	34,5 h	1,5
In order to complete the module, yo	ou have to	fulfil the following	requirements:			
Compulsory Attendance						
Active participation	According	ccording to § 5 para. 3				
Coursework						
Module examination	Usually w	ritten exam (120 m	in), alternatively	v oral exam (30) min)	
Qualification Goals, learning outcon	ne, compe	tences				
quantum chemical method in a comp They acquire a profound understand in the context of quantum chemistry	outer prog ing of qua . They will	ram step by step sta ntum chemical basic be able to perform	arting from the i cs and gain conf the derivation c	nitial ansatz fo idence in hanc of the correspo	lling mathemat	ction. tical formulae
They acquire a profound understand	outer prog ing of quat . They will e able to d d multi-ele general ide lethod for on and Roc	ram step by step sta ntum chemical basic be able to perform esign a correspondi ctron wave function ea, detailed derivation solving the HF equa othaan-Hall equation	arting from the i cs and gain conf the derivation c ing computer pr on of the corres tions ns	nitial ansatz fo idence in hanc of the correspo ogram. ponding equat	or the wavefun lling mathemat anding equation	ction. tical formulae
They acquire a profound understand in the context of quantum chemistry how the equations are solved and ar Contents Molecular orbitals and Hartree-Fock theory (g Self-consistent field m Basis set representation	outer prog ing of quat . They will e able to d d multi-ele general ide tethod for on and Roo -SCF and p	ram step by step sta ntum chemical basid be able to perform esign a correspondi ctron wave function ea, detailed derivation solving the HF equa othaan-Hall equation erformance of corre	arting from the i cs and gain conf the derivation of ing computer pr on of the corres tions ns esponding calcu	nitial ansatz fo idence in hanc of the correspo ogram. ponding equat	or the wavefun lling mathemat anding equation	ction. tical formulae
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Module PQC	Program	nming in Quant	um Chemist	ry	[Modul-ł	(ennnummer]	
Mandatory or elective Module	Elective				<u> </u>		
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	er					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Practical Course "Programming in Quantum Chemistry"	Apr	1 o. 3 (2)	м	3	103,5 h	4,5	
b) Supporting Seminar to a)	S	1 o. 3 (2)	М	1	34,5 h	1,5	
In order to complete the module, you	have to	fulfil the following	requirements:				
Compulsory Attendance	Apr, S						
Active participation	According to § 5 para. 3						
Coursework					0		
Module examination							
Qualification Goals, learning outcome	e, compet	ences					
and critically discuss the Contents Basics of programming Planning and conceptio Implementation of quar	n of a cor						
Compulsory entrance requirements			Module "Princi	iples of Quantu	um Chemistry"		
Recommended participation requirer and/or individual courses of the mod		or the module					
Language(s) of instruction and exami	nation		German or English				
Weight of the module grade in the ov	erall grad	le	Not graded				
Frequency of module offer			Only in the winter term in the semester break				
Reasons for compulsory attendance			According to HochSchG § 26 Para. 2 (7), internship; internship-accompanying upper seminar according to § 5 Para. 5: Discussion of the tasks to be carried out or carried out in the internship with the help of licensed programme on computers within the working group.				
			on computers		king group.		
Person responsible for the module			UnivProf. Dr.		king group.		
	r degree	programs		Jürgen Gauß			



Module CCP	Practic	al Computationa	al Chemistry		[Mo	odul-Kennnummer]	
Mandatory or elective Module	Elective						
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	ter					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Stu	dy Creditpoints	
a) Practical Course Computer Chemistry	APr	2 (1 o. 3)	М	3	103,5	h 4,5	
b) Supporting Seminar to a)	S	2 (1 o. 3)	М	1	34,5 h	n 1,5	
In order to complete the module, ye	ou have to	fulfil the following	ing requirements:				
Compulsory Attendance	APr, S						
Active participation	Accordin	g to § 5 para. 3					
Coursework					0		
Module examination							
Qualification Goals, learning outcor	ne, compe	tences					
methods from the fiel Contents Carrying out 2-4 exemplary experime the fields of AC, OC, PC, KC and/or b	ents in whi iochemistr	ch chemical issues a	are investigated	from a combi	nation of e	experiments from	
Compulsory entrance requirements							
Recommended participation require and/or individual courses of the mo		or the module	Modules "Cont and "Principles			ntum Chemistry" y"	
Language(s) of instruction and exan	nination		German or English				
Weight of the module grade in the	overall gra	de	Not graded				
Frequency of module offer			Only in the summer term in the lecture-free period				
Reasons for compulsory attendance			According to HochSchG § 26 Para. 2 (7), practical course; internship-accompanying upper seminar according to § 5 Para. 5: Discussion of the tasks to be carried out or carried out in the internship with the help of licensed programme on computers within the working group.				
Person responsible for the module			UnivProf. Dr.	Jürgen Gauß			
Transferability of the module to oth	er degree	programs	Master of Scier	nce Chemistry			
Other			Block practical	course			
INI							

Module MTTC	Conten	nporary Topics o	of Theoretica	l Chemistry	[Modul-H	(ennnummer]	
Mandatory or elective Module	Elective				<u>4</u>		
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	er					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoint	
a) Lecture "Contemporary Topics of Theoretical Chemistry"	L	2 (1 o. 3)	М	3	103,5 h	4,5	
b) Supporting exercise to a)	E	2 (1 o. 3)	М	1	34,5 h	1,5	
In order to complete the module, yo	u have to	fulfil the following	requirements:				
Compulsory Attendance							
Active participation	According	According to § 5 para. 3					
Coursework							
Module examination	Usually w	ritten exam (120 m	in), alternativel	y oral exam (3	0 min)		
Qualification Goals, learning outcom	e. compe	tences					
They have developed t Chemistry in Practice". Contents Advanced quantum chemic Theoretical description of many-parti	al method	s	0		the module "Co	omputational	
Compulsory entrance requirements	cie system			orrelation			
Recommended participation require and/or individual courses of the mod		or the module					
Language(s) of instruction and exam	ination		German or English				
Weight of the module grade in the o	verall grad	de	Not graded				
Frequency of module offer			Only in the summer term				
Reasons for compulsory attendance							
Person responsible for the module			UnivProf. Dr. Jürgen Gauß				
Transferability of the module to oth	er degree	programs	Master of Scier	nce Chemistry			
Other							

Module KIWE	Artificia	I intelligence in	drug discov	ery and	[Modul-I	Kennnummer]
	develop	oment				
Mandatory or elective Module	Elective					
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration (according to course plan)	1 Semest	er				
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints
a) Lecture "Al in healthcare"	L	2 (1 o. 3)	М	3	103,5 h	4,5
b) Supporting exercise to a)	E	2 (1 o. 3)	М	1	34,5 h	1,5
In order to complete the module, you	u have to	fulfil the following	requirements:			
Compulsory Attendance						
Active participation	According	to § 5 para. 3			0	
Coursework						
Module examination	Usually w and b)	ritten exam (120 m	in), alternatively	y oral exam (3	0 min) on the c	ontents of a)
Qualification Goals, learning outcom	e, compet	ences				
3. Methods f b) Molecular docking 1. Classical a 2. Al approac c) Molecular dynamics 1. Classical a	e in drug de e data on s Al tools in ed drug de rug design and design and design drug des and funct d nucleic s mutations or structu pproaches ches in mo pproaches ches in mo poros	lesign in academic of small molecule opti the design of nove sign (CADD) and Al design (Virtual scre the (supervised/unsup ign ion of biomolecules	or industrial sett mization metho I compounds. in CADD ening, pharmac pervised learning s ction ng (virtual scree fusion models, i mics (binding fre	tings. ods and proces cophore screer g, generative r ening, cross-do flow-based mo ee energy calc	ning, Structure- modeling such ocking) odels) ulations)	-
Compulsory entrance requirements						
Recommended participation required and/or individual courses of the mod		or the module				
Language(s) of instruction and exami	nation		English			
Weight of the module grade in the or	verall grad	le	Not graded			
Frequency of module offer			Every winter te	erm		
Reasons for compulsory attendance						

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Person responsible for the module	UnivProf. Dr. Paul Czodrowski
Transferability of the module to other degree programs	
Other	

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Remarks

Depending on the type of course, different conversion factors are used to calculate a certain number of ECTS.

In general:

1 ECTS corresponds to 30h total workload (time hours),

1 corresponds to 10.5h attendance time per semester (14 weeks à 0.75h)

xe"

Contact time (SWH)	1	2	3	4
Total attendance time	10,5h	21h	31 <i>,</i> 5h	42h

Lectures and/or Exercises

A factor of 1.5 is applied, i.e. 2 lectures or exercises correspond to 3 ECTS.

4,5 ECTS	3 (e.g. 2L+1E), 31,5h attendance time, 103.5h self-study, 135h total workload
6,0 ECTS	4 (e.g. 3L+1E), 42h attendance time, 138h self-study, 180h total workload
7,5 ECTS	5 (e.g. 3L+2E), 52,5h attendance time, 172,5h self-study, 225h total workload

Practical Courses

A factor of 0.50 or 0.75 or 1.00 is applied, depending on the extent of preparation and follow-up, e.g. with reports, ...

6,0 ECTS	Factor 0,50	12, 126h attendance time, 54h self-study, 180h total workload
		e.g. 10 weeks of 12,6h
7,5 ECTS	Factor 0,50	15, 157,5h attendance time, 67,5h self-study, 225h total workload
		e.g. 10 weeks of 15h
7,5 ECTS	Factor 0,75	10, 105h attendance time, 120h self-study, 225h total workload
		e.g. 10 weeks of 10,5h
6,0 ECTS	Factor 1,00	6, 63h attendance time, 117h self-study, 180h total workload
		e.g. 10 weeks of 6,3h

Seminars

A factor of 1.0 or 1.5 is applied, depending on the amount of preparation and follow-up, e.g. with lectures, new learning material, ...

1,0 ECTS	Factor 1,0	1, 10,5h attendance time, 19,5h self-study, 30h total workload
2,0 ECTS	Factor 1,0	2, 21h attendance time, 39h self-study, 60h total workload
1,5 ECTS	Factor 1,5	1, 10,5h attendance time, 34,5h self-study, 45h total workload
3,0 ECTS	Factor 1,5	2, 21h attendance time, 69h self-study, 90h total workload

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Abbreviations

Abbreviation	Meaning	7
BMC	Biomedical Chemistry	-
e.g.	For example	
ECTS / CP(LP)		
IUPAC		
	International Union of Pure and Applied Chemistry	
SWH(SWS)	Hours per Semester Week	
S E	Seminar Exercise	
Apr	Advanced Practical Course	
	Lecture	
information without buck		